

# Membrane Biochemistry

Lectures by

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Sunday, November 28, 2010

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## Further reading

- [Chemiosmotic coupling: The cost of living](#). By Peter Rich. (.pdf file, 80 kb)
- [Power for Life](#). Review of Nick Lane's book "Power Sex Suicide...." (.pdf file, 416 kb)
- [N,K-ATPase](#). Page of Mark Hilge at Protein Biophysics, Nijmegen
- [ATP Synthase](#). Group Pages of John Walker at the MRC Mitochondrial Biology Unit, Cambridge

## Animations

- [The pump cycle of Na,K-ATPase](#). By Mark Hilge at Protein Biophysics, Nijmegen
- [Animation. From Light to ATP](#). By O. Fritsche and W. Junge, University of Osnabruck. (.avi file, 17.7 mb)
- [Molecular animations of ATP synthase](#). From the research group of John Walker at the MRC Mitochondrial Biology Unit, Cambridge
- [Animation. Powering the Cell: Mitochondria](#). From BioVisions at Harvard University

## Relevant Nobel prizes

- 1906 Nobel Prize in Physiology or Medicine to Camillo Golgi and Santiago Ramón y Cajal
- 1974 Nobel Prize in Physiology or Medicine to Albert Claude, Christian de Duve and George E. Palade
- 1978 Nobel Prize in Chemistry to Peter Mitchell
- 1988 Nobel Prize in Chemistry to Johann Deisenhofer, Robert Huber and Hartmut Michel
- 1997 Nobel Prize in Chemistry to Paul D. Boyer, John E. Walker and Jens C. Skou
- 1999 Nobel Prize in Physiology or Medicine to Günter Blobel





# Lectures in Membrane Biochemistry

- [The endomembrane system - endocytosis and exocytosis \(Acrobat, .pdf file\)](#)
- [The endomembrane system - vesicular transport and protein trafficking \(Acrobat, .pdf file\)](#)
- [Transport across membranes 1 - Proteins \(Acrobat, .pdf file\)](#)
- [Transport across membranes 2 - Small molecules and ions \(Acrobat, .pdf file\)](#)
- [Mitochondria and chloroplasts - analysis of compartments \(Acrobat, .pdf file\)](#)
- [Bioenergetics \(Acrobat, .pdf file\)](#)
- [The respiratory chain and oxidative phosphorylation \(Acrobat, .pdf file\)](#)
- [Oxidative phosphorylation, respiratory control and the chemiosmotic hypothesis \(Acrobat, .pdf file\)](#)
- [ATP Synthase - Coupling ATPase \(Acrobat, .pdf file\)](#)

Course web pages

Membrane Biochemistry web pages

General reference

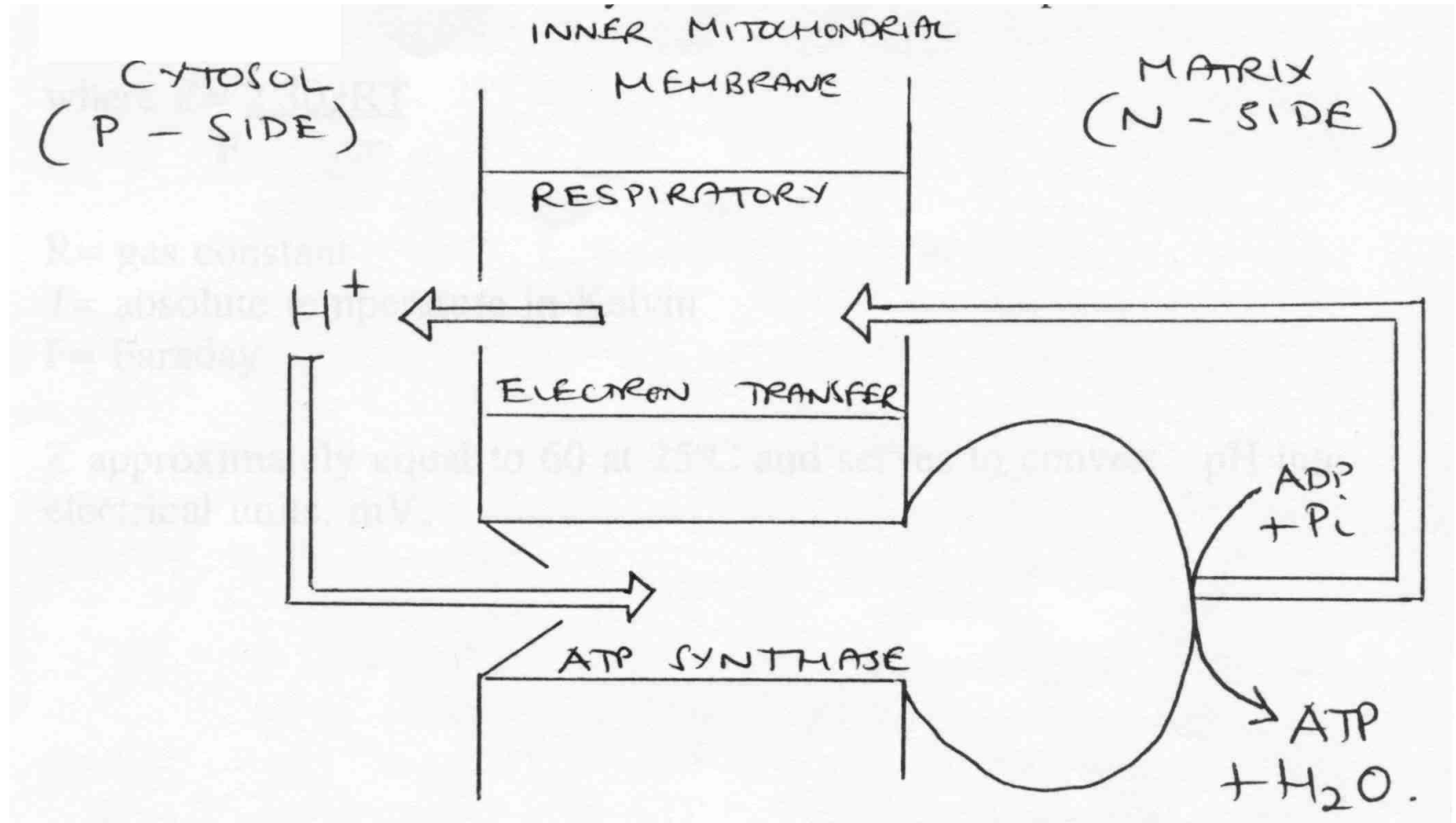
[Cell and Molecular Biology: Concepts and Experiments](#)  
Gerald Karp. Fifth Edition 2008. John Wiley & Sons Inc.

# ATP Synthase – Coupling ATPase

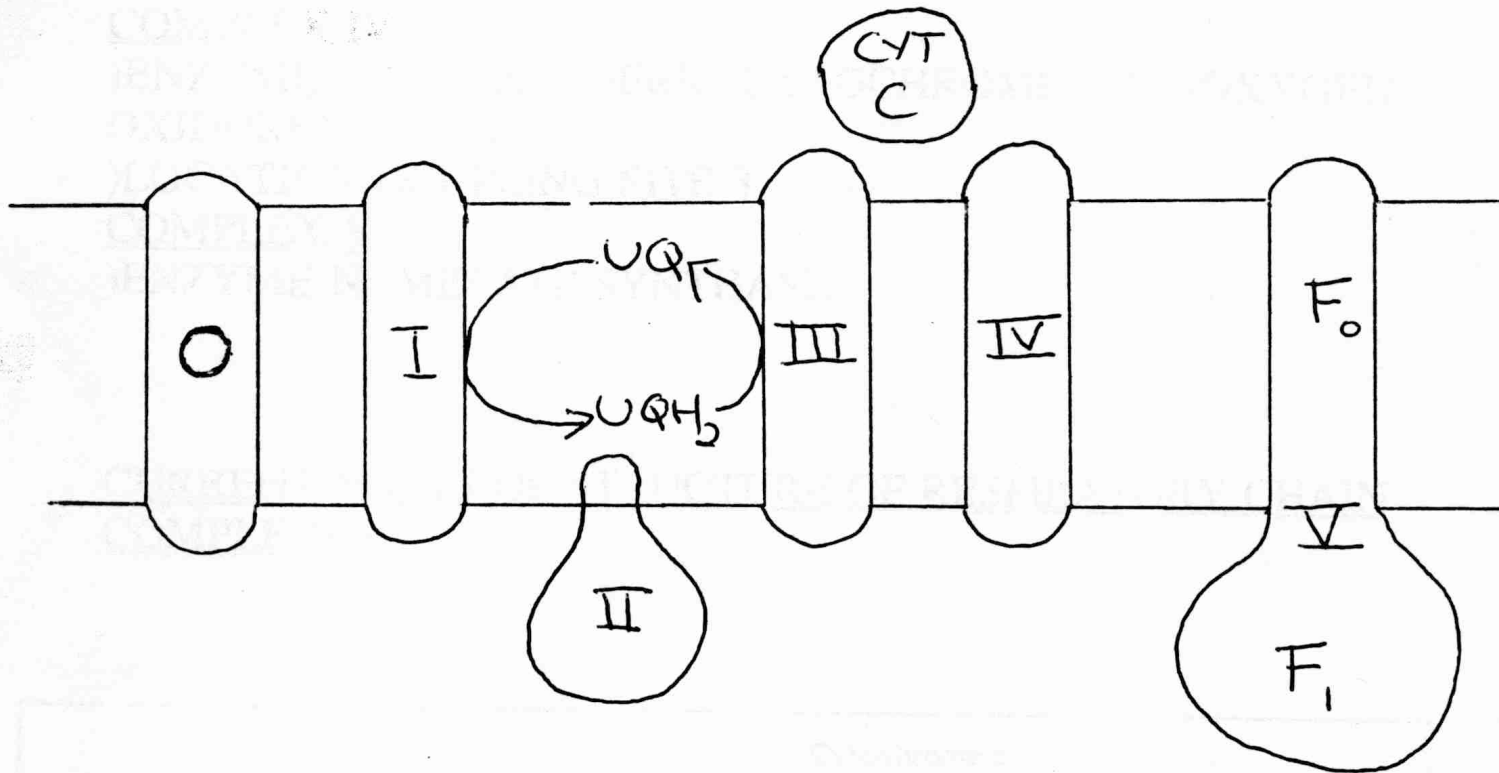


## Transmembrane Proton gradient

Energy transduction occurs by means of a proton circuit through the insulating, coupling membrane and between the two bulk aqueous phases (the matrix and the intermembrane space/cytosol in mitochondria).

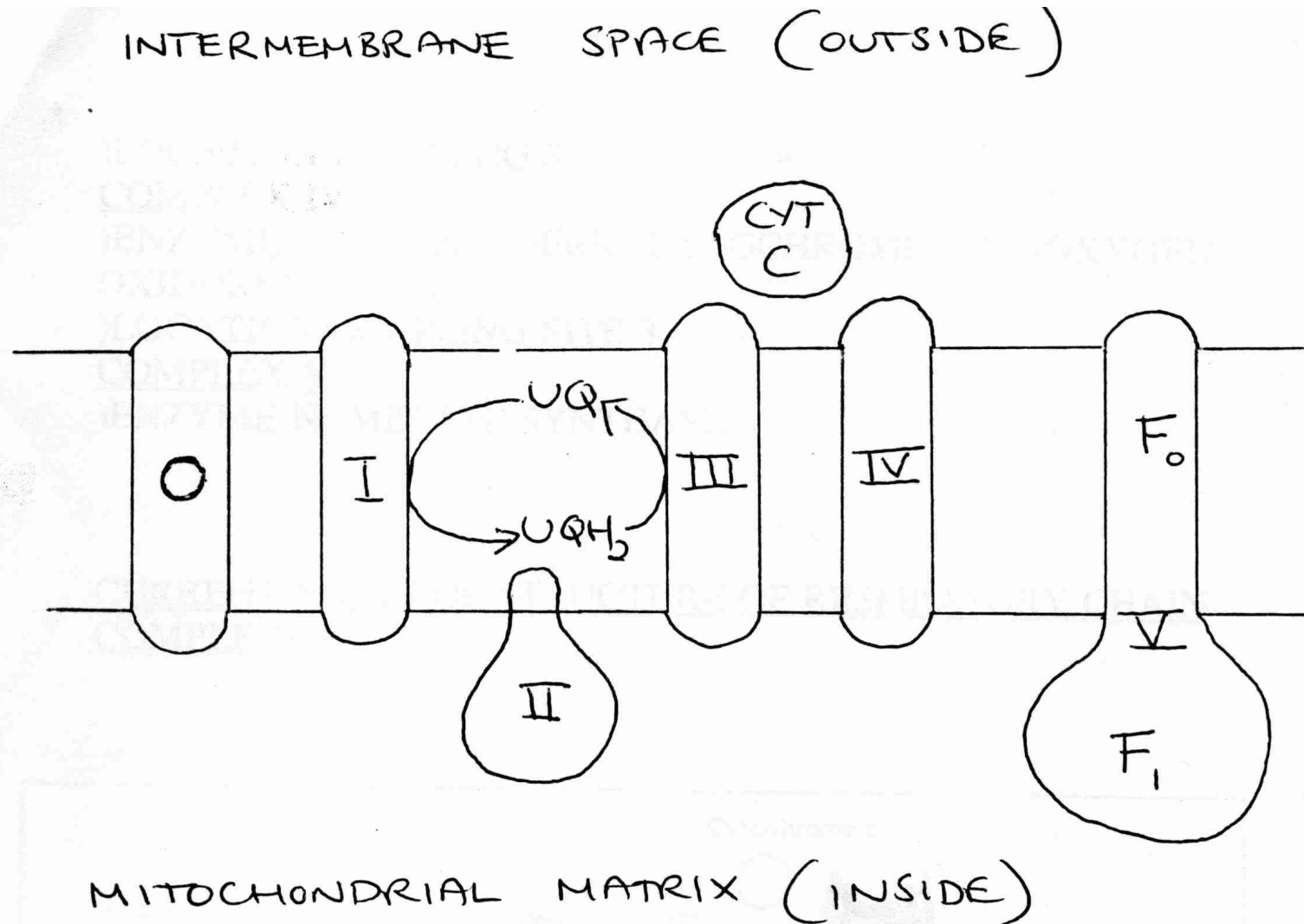


INTERMEMBRANE SPACE (OUTSIDE)



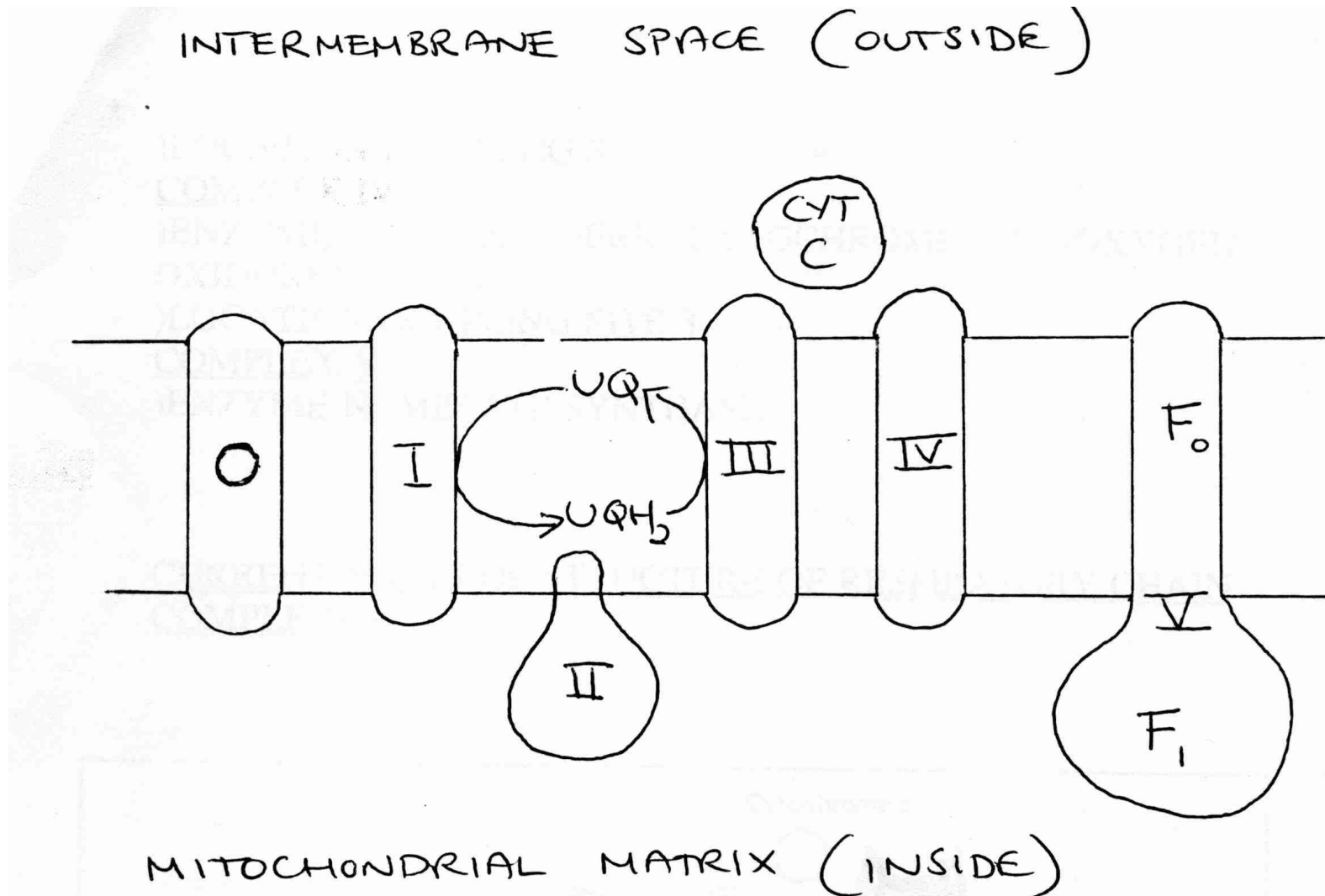
MITOCHONDRIAL MATRIX (INSIDE)

# COMPLEX V



## COMPLEX V

ENZYME NAME: COUPLING ATPase; ATP SYNTHASE

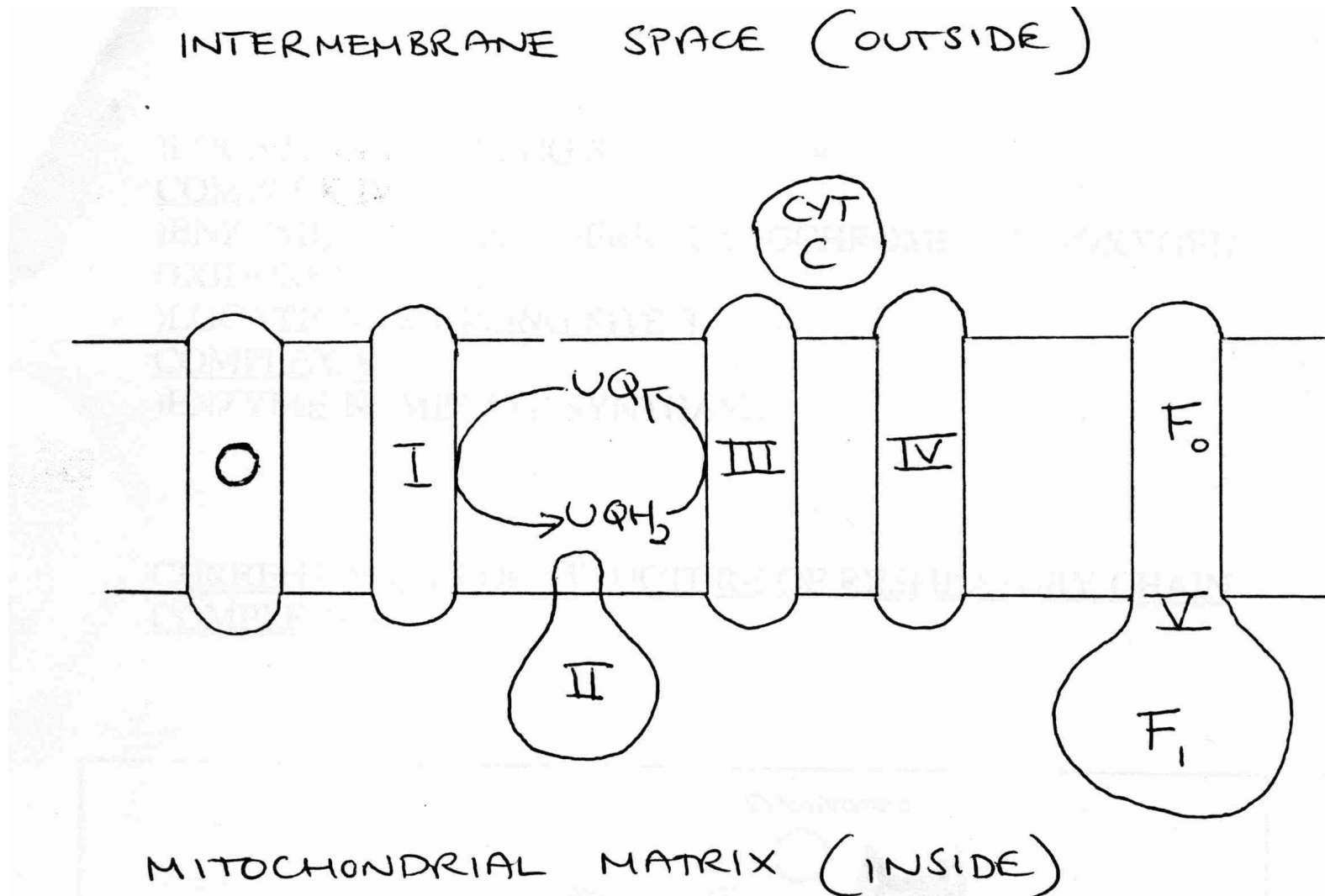




## COMPLEX V

ENZYME NAME: COUPLING ATPase; ATP SYNTHASE

F-ATPase

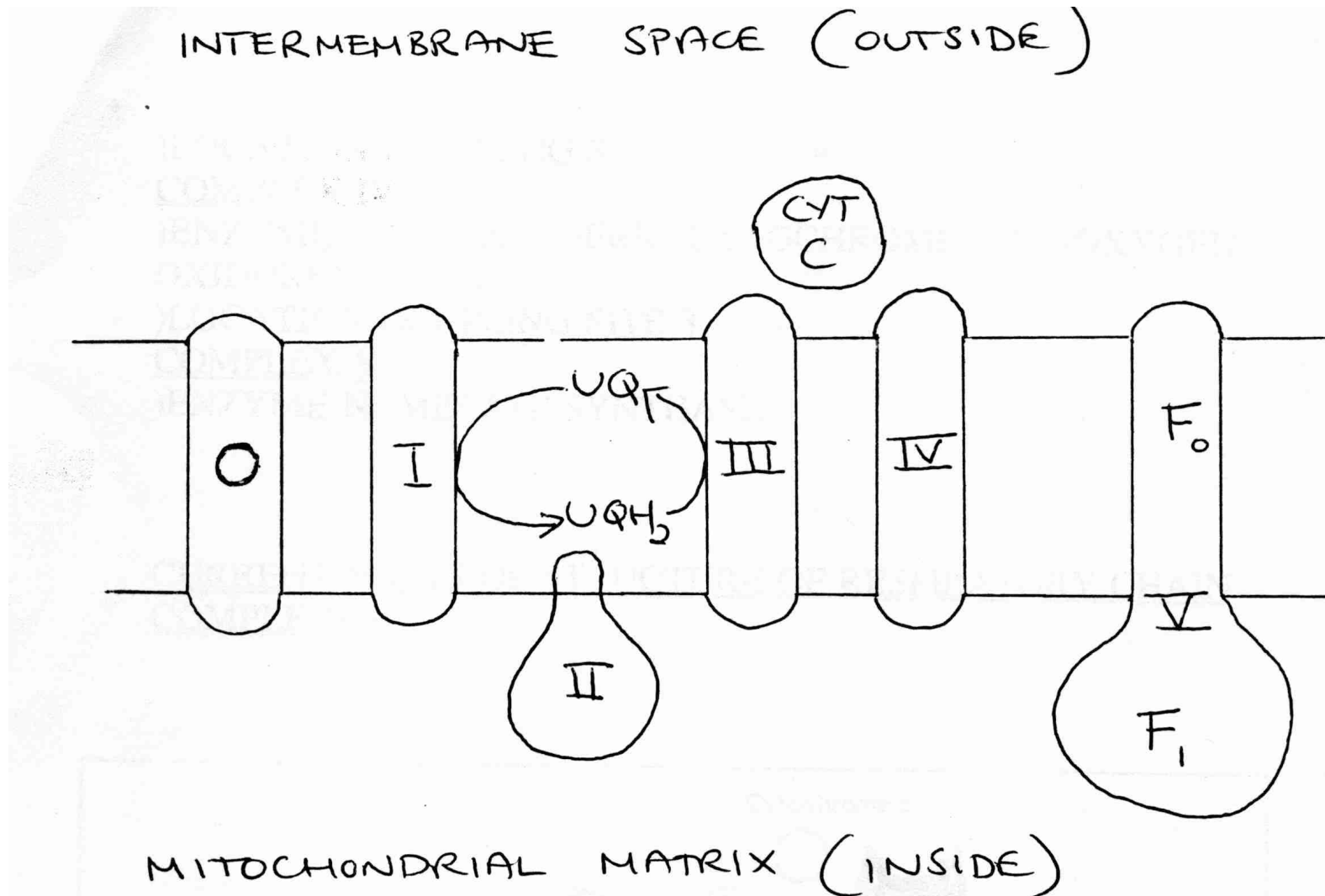


## COMPLEX V

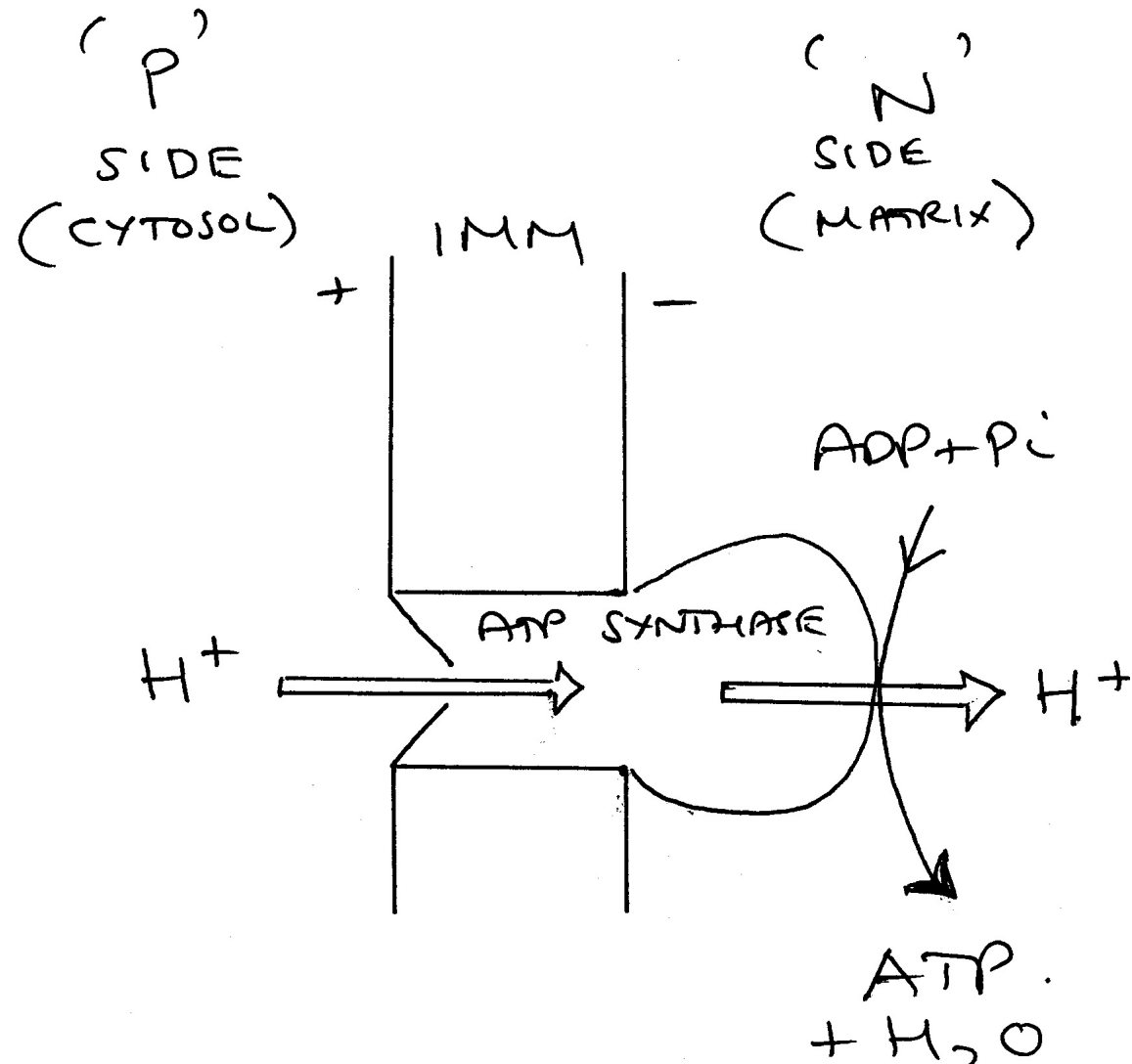
ENZYME NAME: COUPLING ATPase; ATP SYNTHASE

F-ATPase

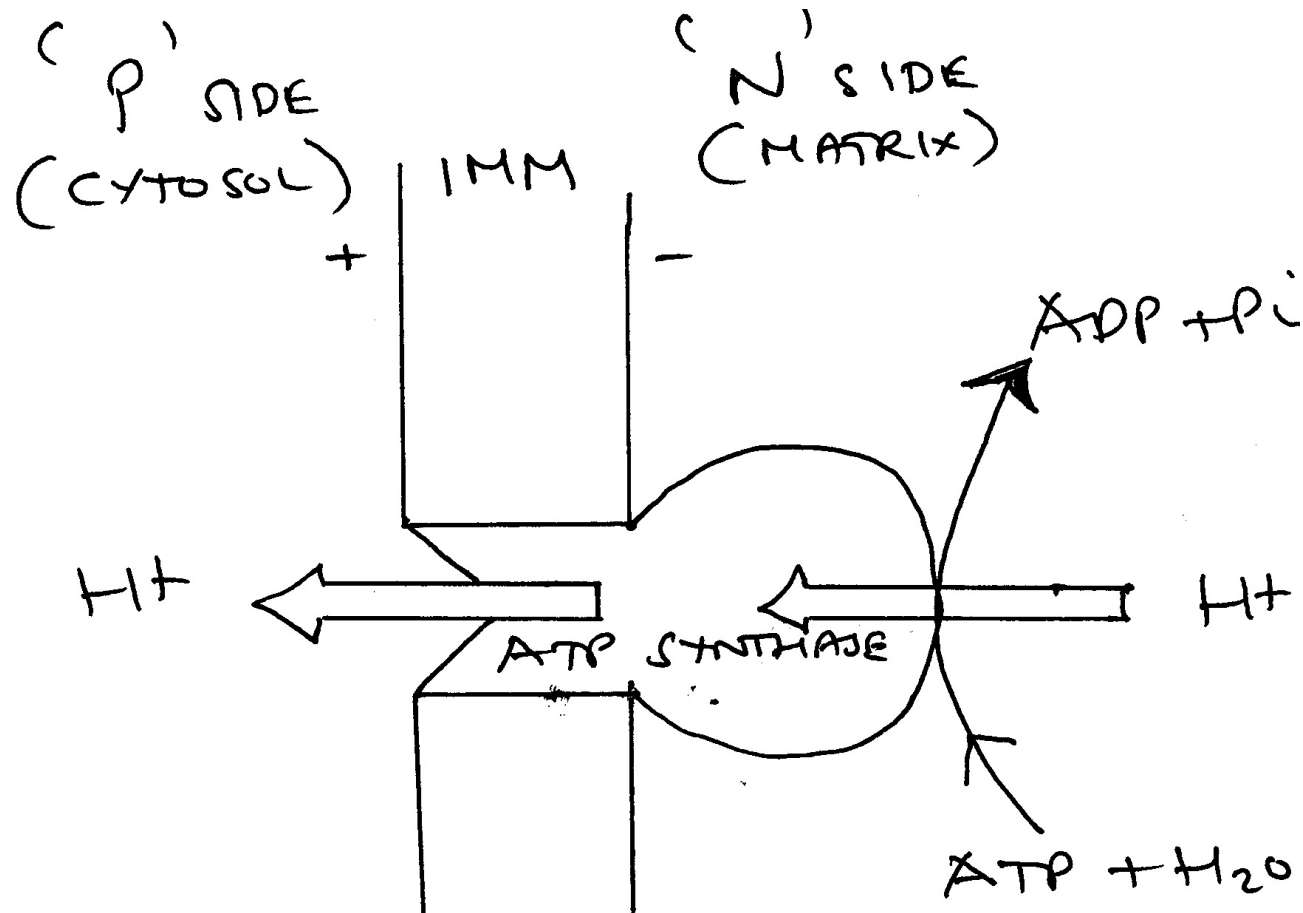
F<sub>1</sub>-F<sub>0</sub> ATPase



ATP synthase can transport protons across the membrane in one direction down the concentration ( $\Delta \text{pH}$ ) and charge ( $\Delta \psi$ ) gradient, using the energy for ATP synthesis.



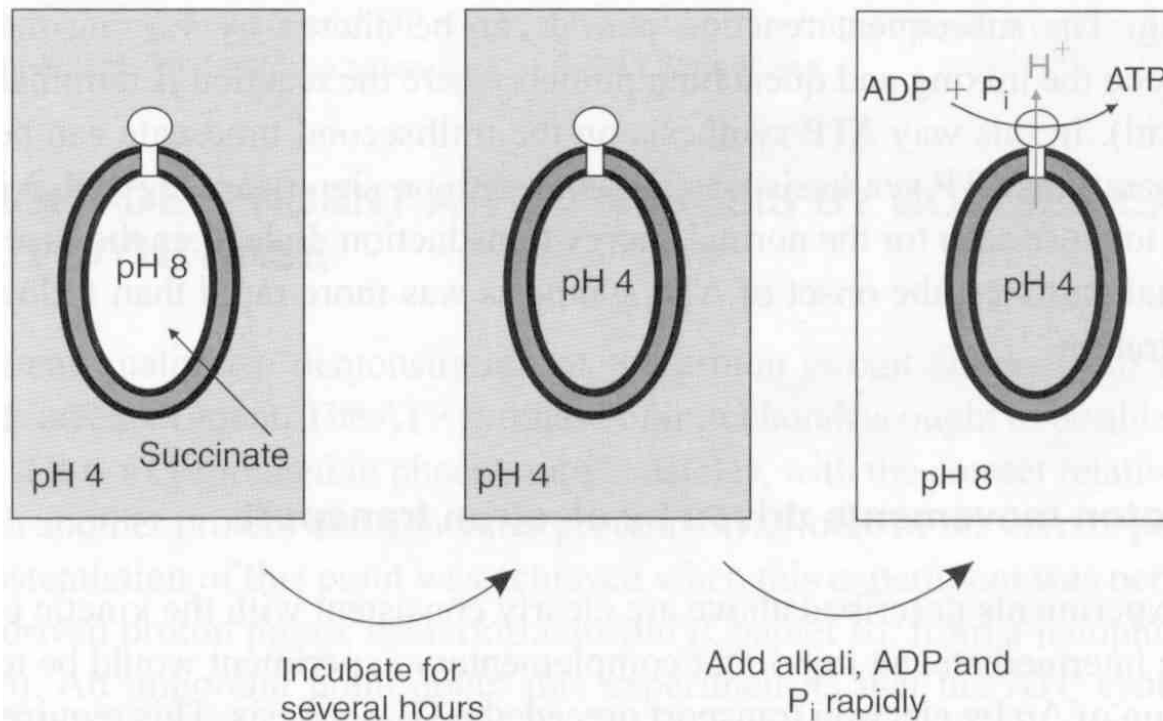
Alternatively the ATP synthase can use the energy from ATP hydrolysis to pump protons in the opposite direction (active transport against the concentration and charge gradient).





# Evidence for $\Delta\text{pH}$ -driven ATP synthesis

1. The 'acid-bath experiment'.  $\Delta\text{pH}$  driving ATP synthesis was demonstrated in 1966 by Jagendorf and Uribe, using thylakoid membranes of chloroplasts.

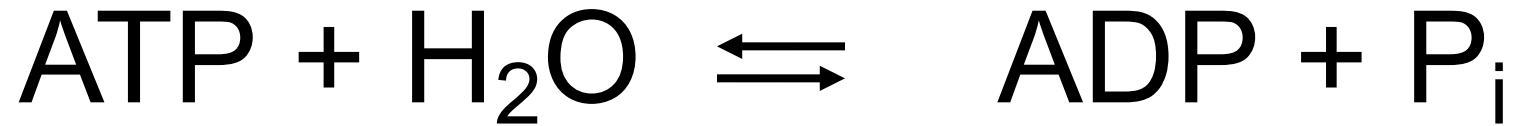


**Figure 4.15 The 'acid bath' experiment: a  $\Delta\text{pH}$  can generate ATP.**

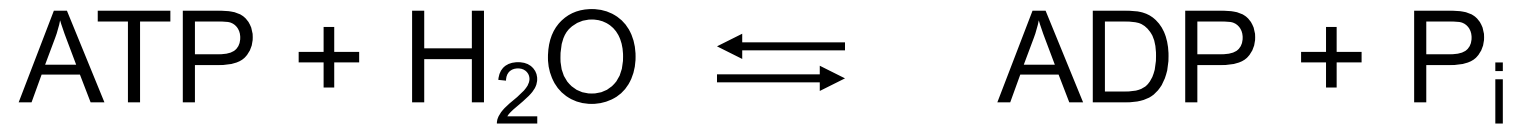
Thylakoid membranes were incubated in the dark at pH 4 in the presence of electron transport inhibitors in a medium containing succinate, which slowly permeated into the thylakoid space liberating protons and lowering the internal pH to about 4. The external pH was then suddenly raised to 8, creating a  $\Delta\text{pH}$  of 4 units across the membrane. Traditionally  $\text{H}^+$  efflux through the ATP synthase has been regarded as charge compensated by  $\text{Cl}^-$  efflux and/or  $\text{Mg}^{2+}$  influx. A  $\Delta\psi$  may also be induced, see text. ADP and  $\text{P}_i$  were simultaneously added and proton efflux through the ATP synthase led to the synthesis of about 100 mol of ATP per mol of synthase. Protonophores such as FCCP inhibited the ATP production.

# Coupling ATPase

# Coupling ATPase



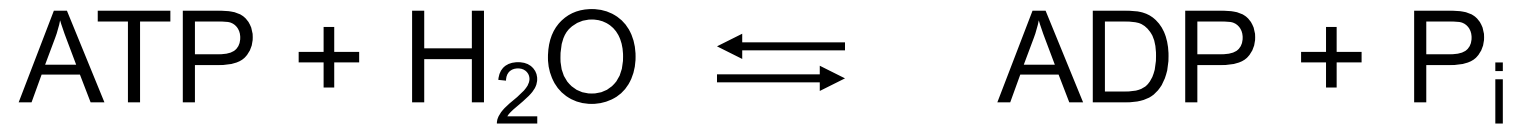
# Coupling ATPase



Also known as “ATP Synthase”



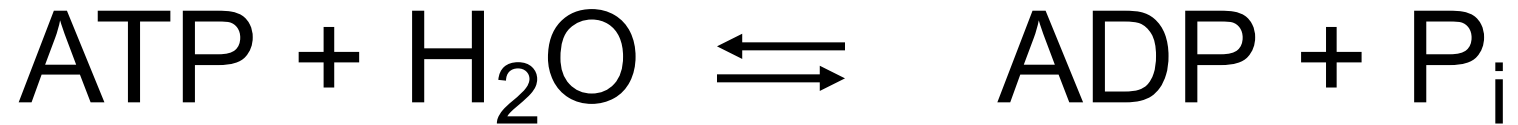
# Coupling ATPase



Also known as “ATP Synthase”

One of the class of F-ATPases (c.f. V-ATPases; P-ATPases)

# Coupling ATPase

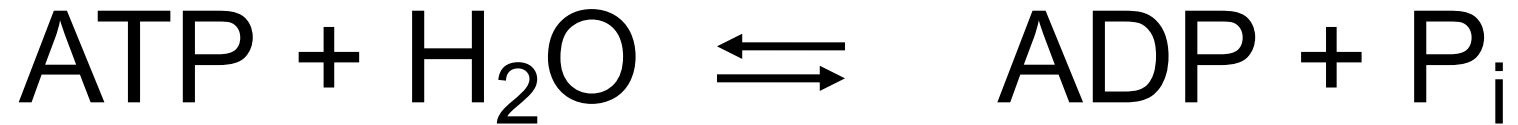


Also known as “ATP Synthase”

One of the class of F-ATPases (c.f. V-ATPases; P-ATPases)

-  $\text{F}_1$

# Coupling ATPase



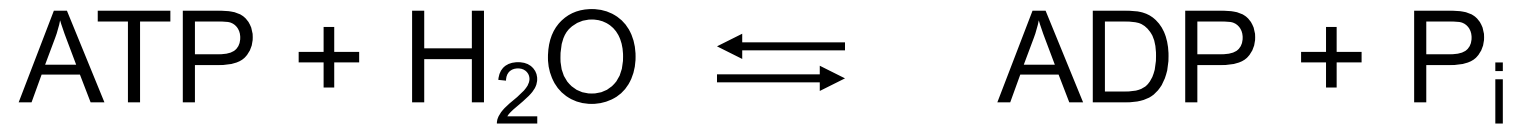
Also known as “ATP Synthase”

One of the class of F-ATPases (c.f. V-ATPases; P-ATPases)

-  $F_1$

-  $F_o$

# Coupling ATPase



Also known as “ATP Synthase”

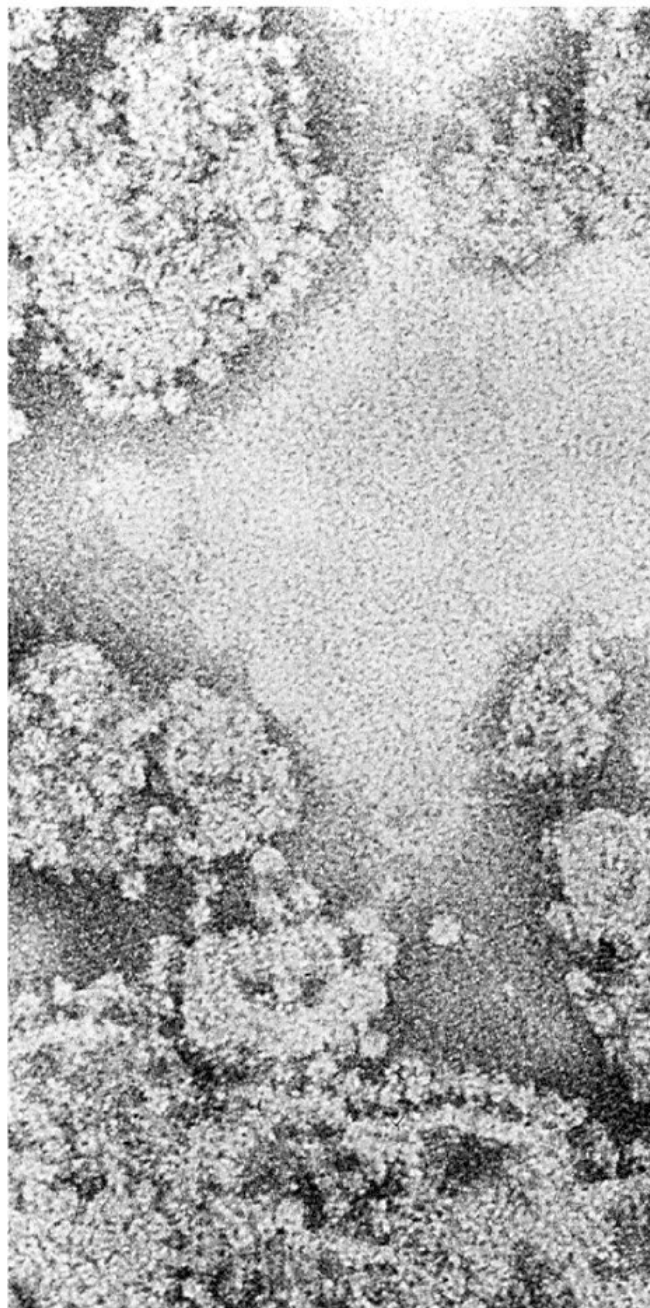
Also known previously as respiratory “complex V”

One of the class of F-ATPases (c.f. V-ATPases; P-ATPases)

-  $F_1$

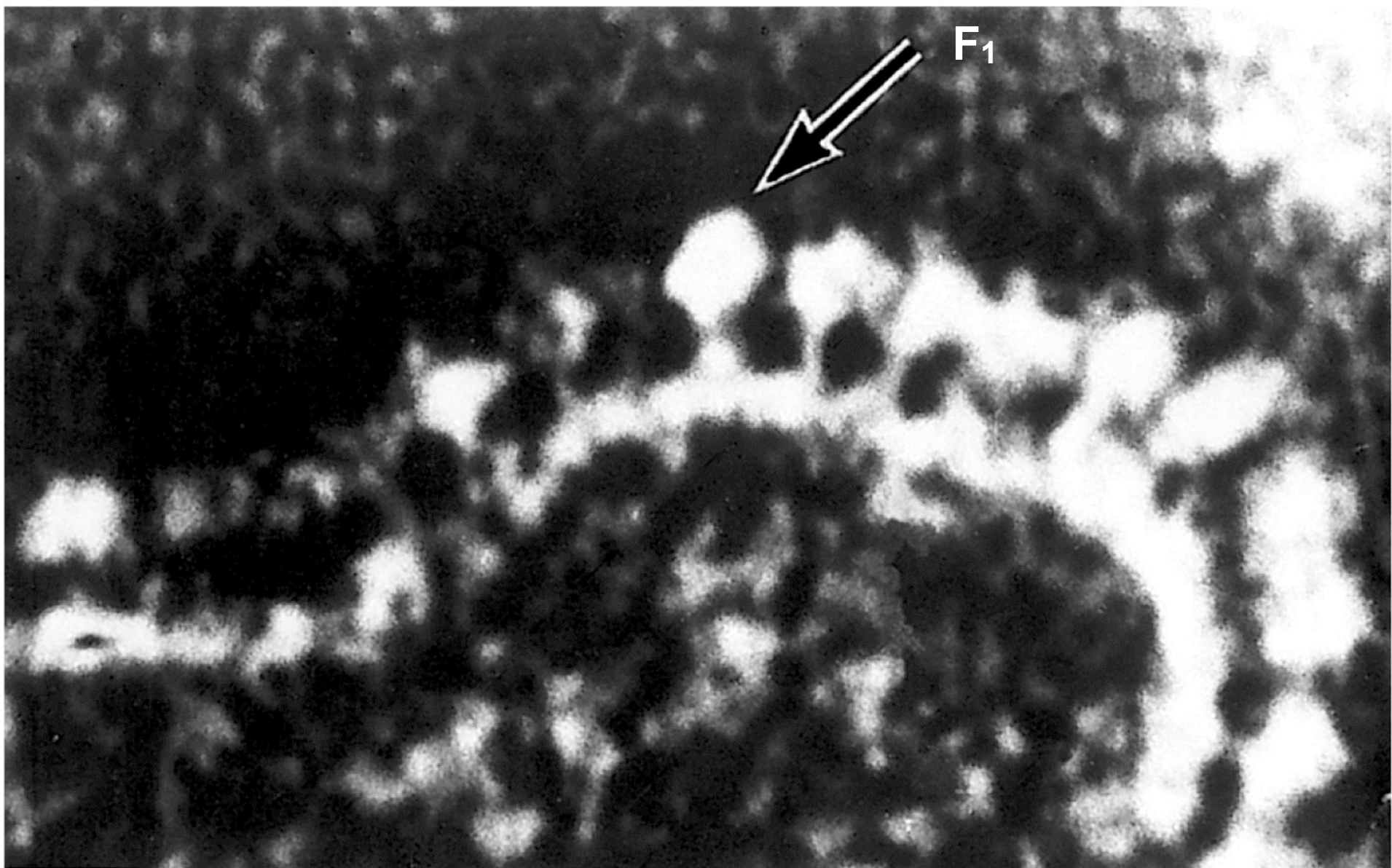
-  $F_o$

$F_1$



**50 nm**

**Figure 5-25a Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)**



10 nm

Figure 5-21 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)



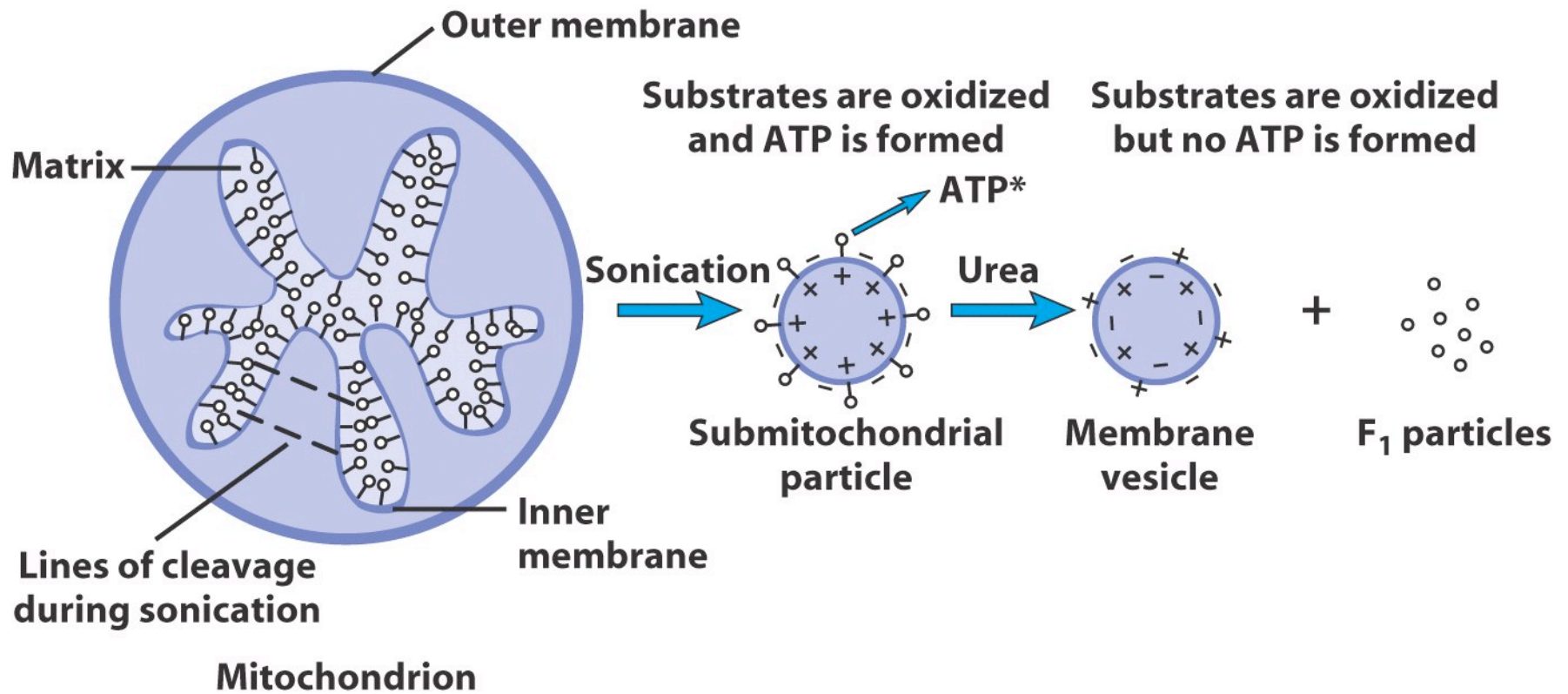
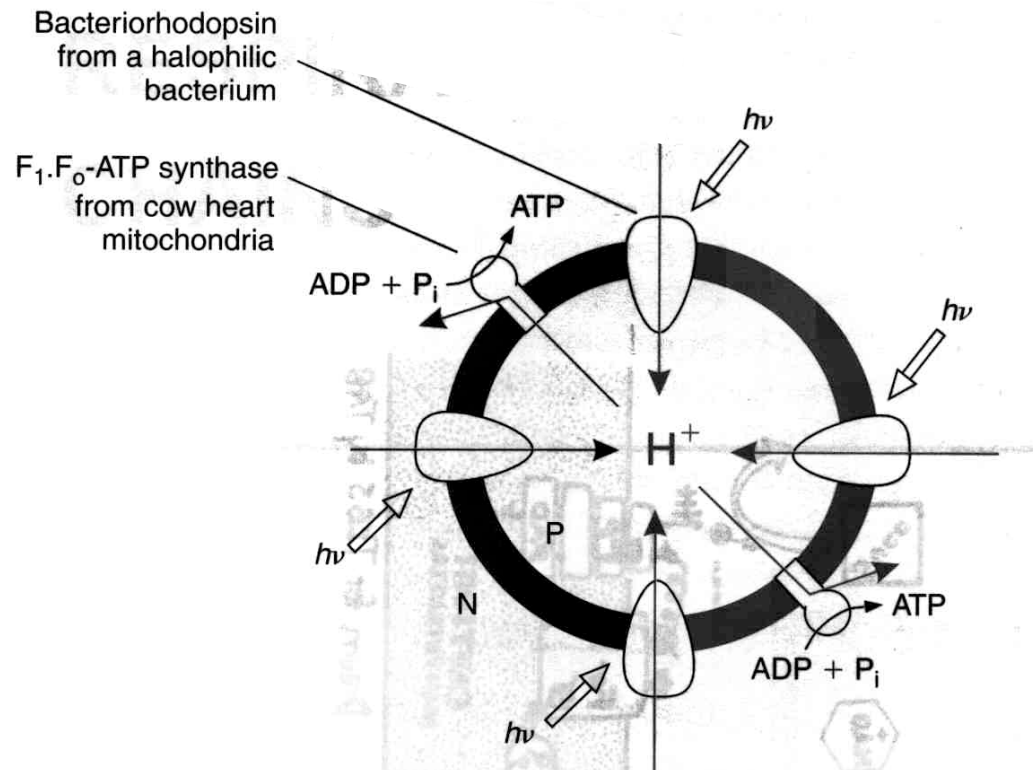


Figure 5-25b Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)

# Evidence for $\Delta pH$ -driven ATP synthesis

2. Experiment of Racker and co-workers, demonstrating light-driven ATP synthesis in lipid vesicles containing bacteriorhodopsin (a light-drive proton pump) and ATP synthase from mitochondria from beef heart



**Figure 4.16 A proton circuit between a light-driven proton pump (bacteriorhodopsin) and ATP synthase from mitochondria.**

The establishment of the proton circuit depends on the majority of the bacteriorhodopsin molecules adopting (for poorly understood reasons) the orientation in which they pump protons inward. Similarly, the ATP synthase had to incorporate predominantly with the topology shown. Opposite orientations of both bacteriorhodopsin and ATP synthase would in principle also have permitted an  $H^+$  circuit, in the opposite direction, to be established. In practice, this would have meant that added ADP and  $P_i$  (both membrane impermeant) would not have been able to reach the active site of the ATP synthase.

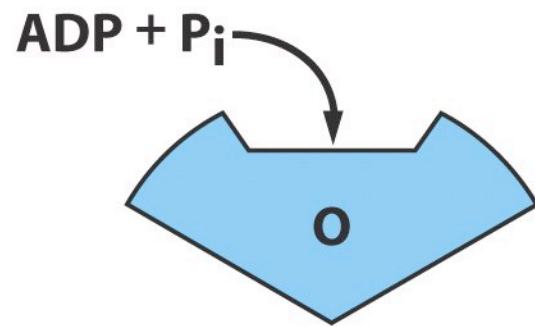


Figure 5-27a part 1 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)

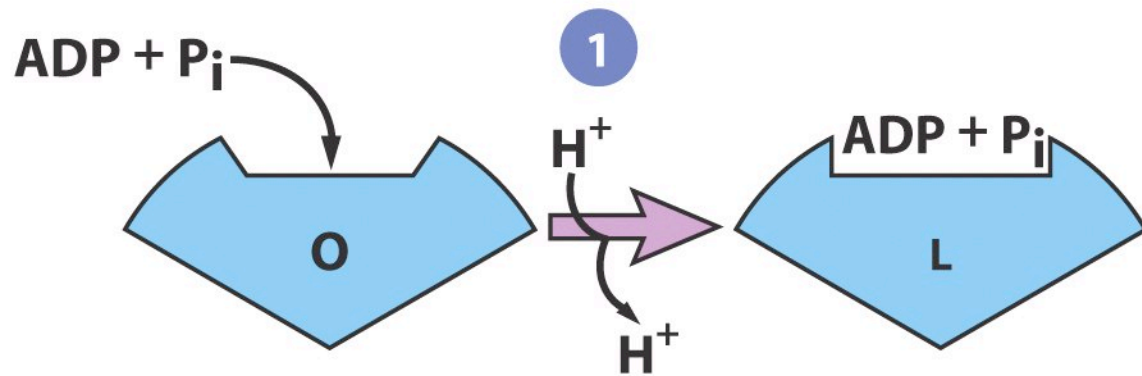


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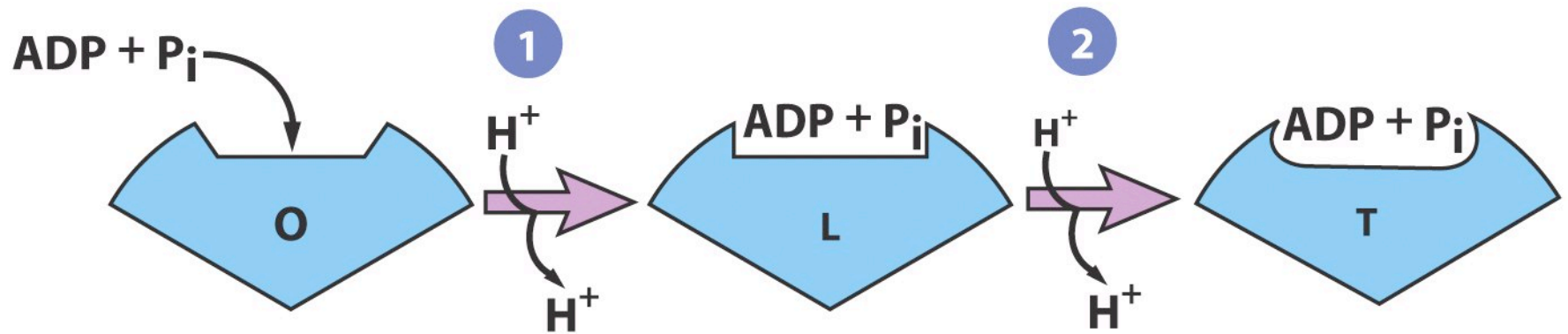


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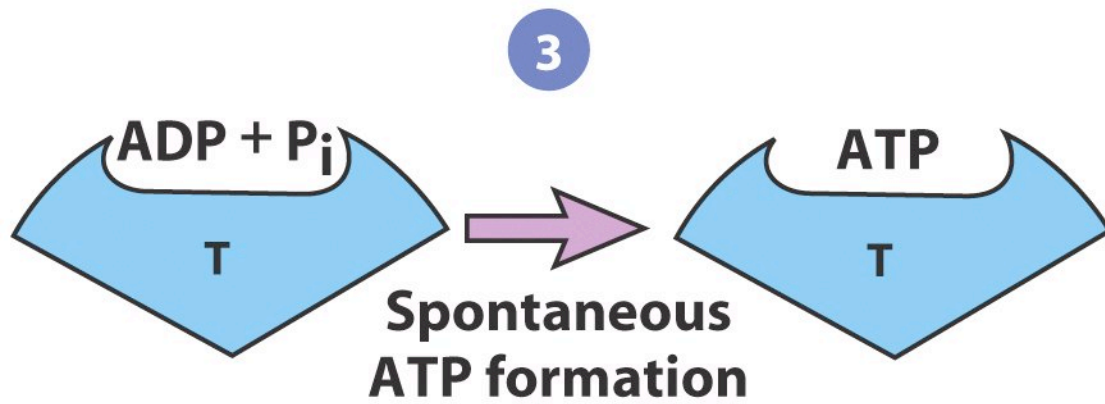


Figure 5-27a part 2 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)

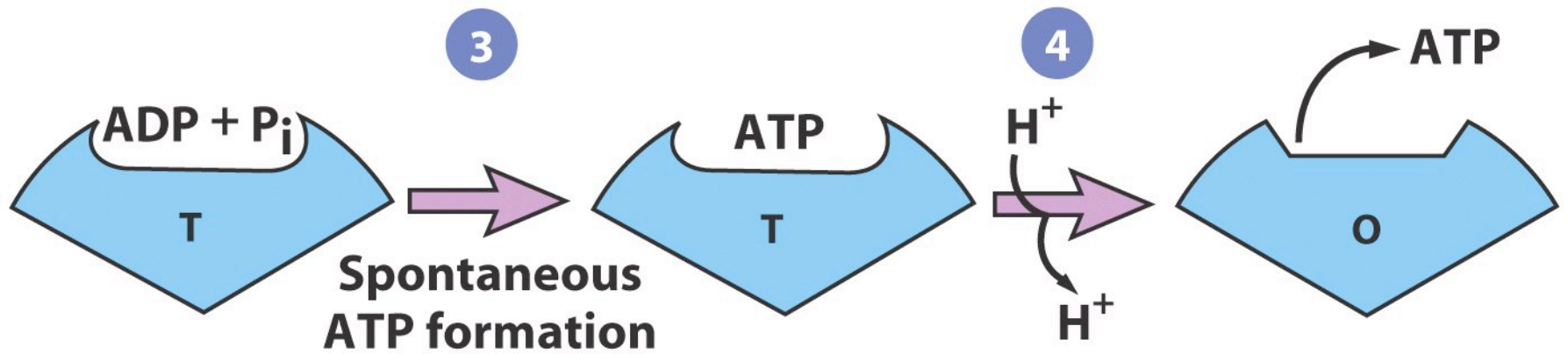


Figure 5-27a part 2 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)



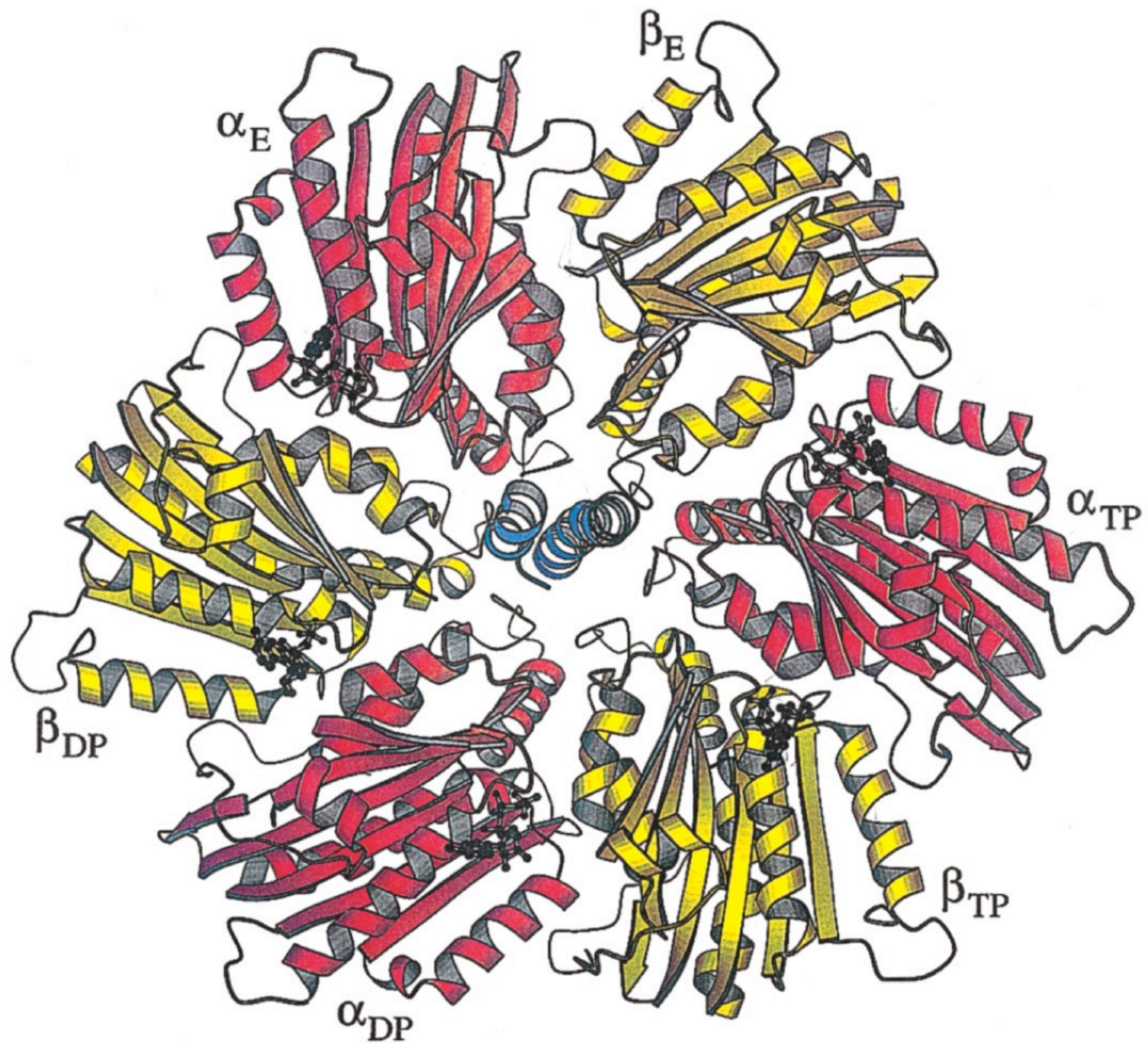


Figure 5-26b Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)

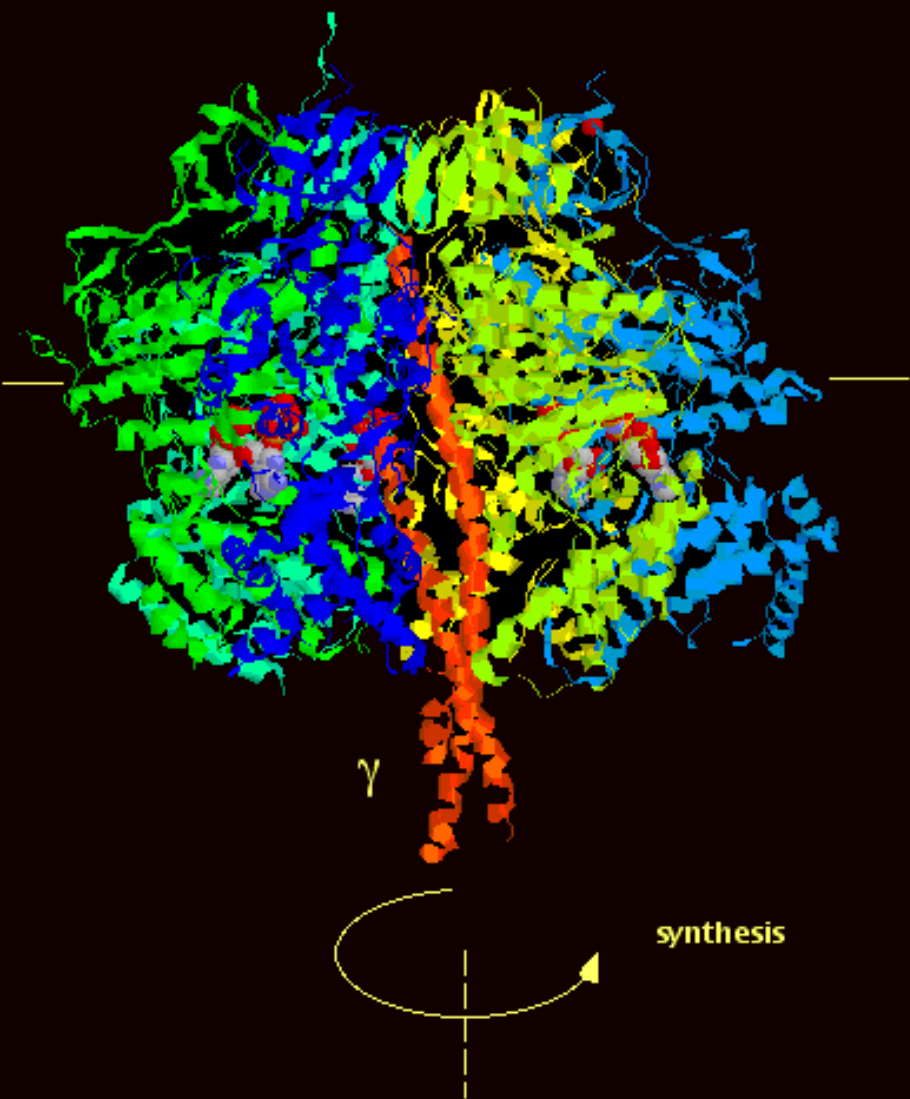


Figure from: ALLEN, J F (2002) Photosynthesis of ATP - Electrons, Proton Pumps, Rotors, and Poise. Cell 110, 273–276



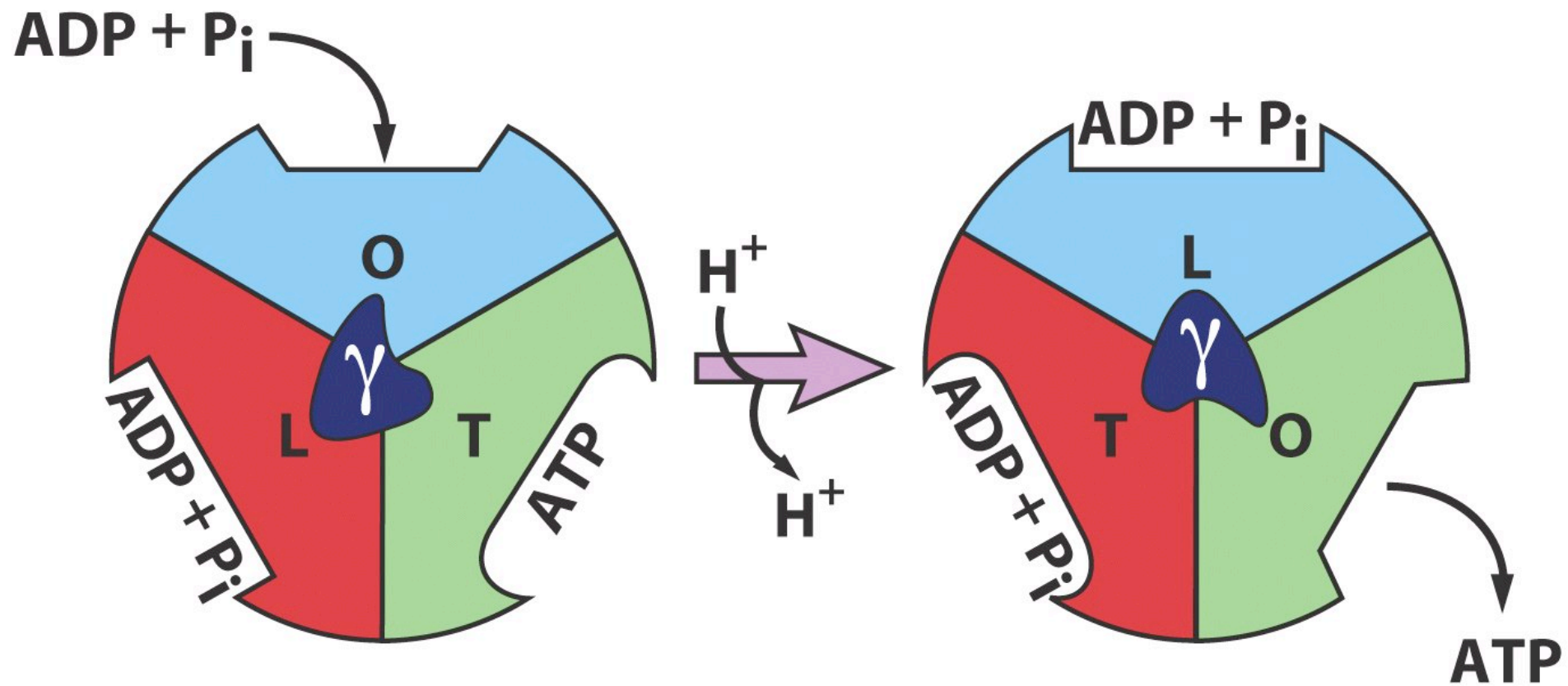


Figure 5-27b part 1 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)



**Spontaneous  
ATP formation  
at red-colored  
catalytic site**

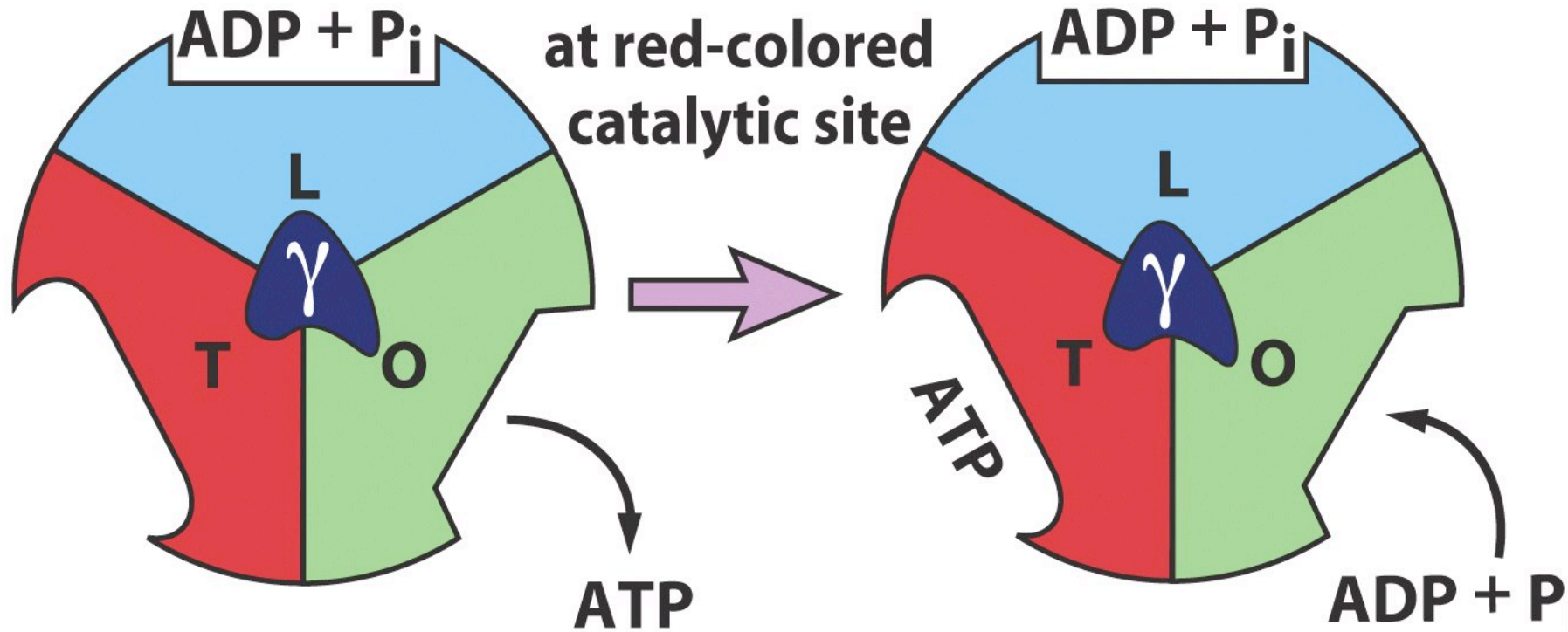


Figure 5-27b part 2 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)

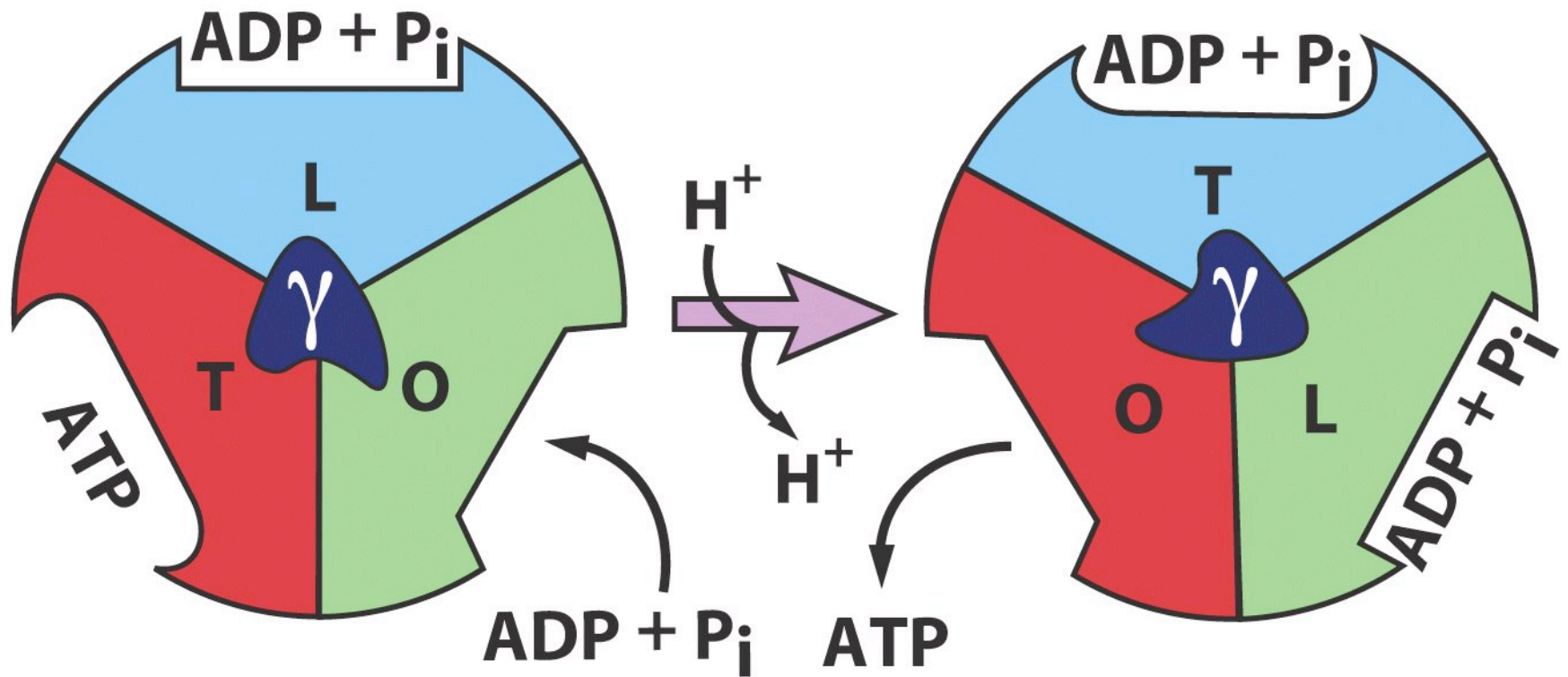


Figure 5-27b part 3 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)

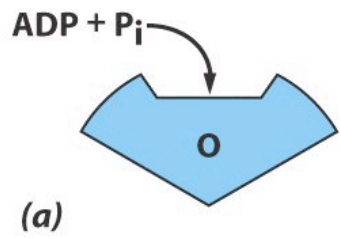


Figure 5-27 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)



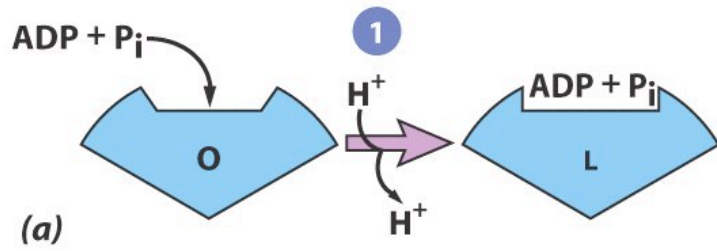


Figure 5-27 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)

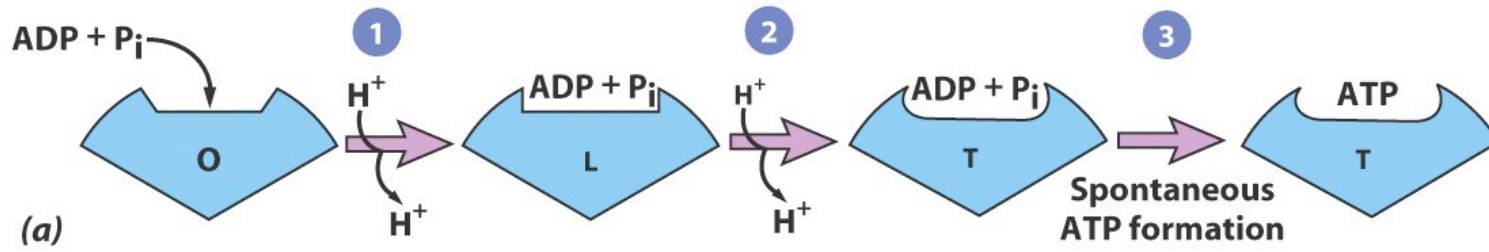


Figure 5-27 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)

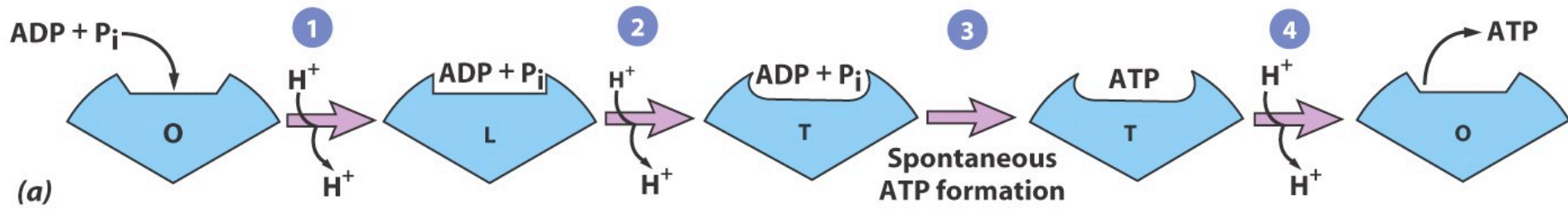


Figure 5-27 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)



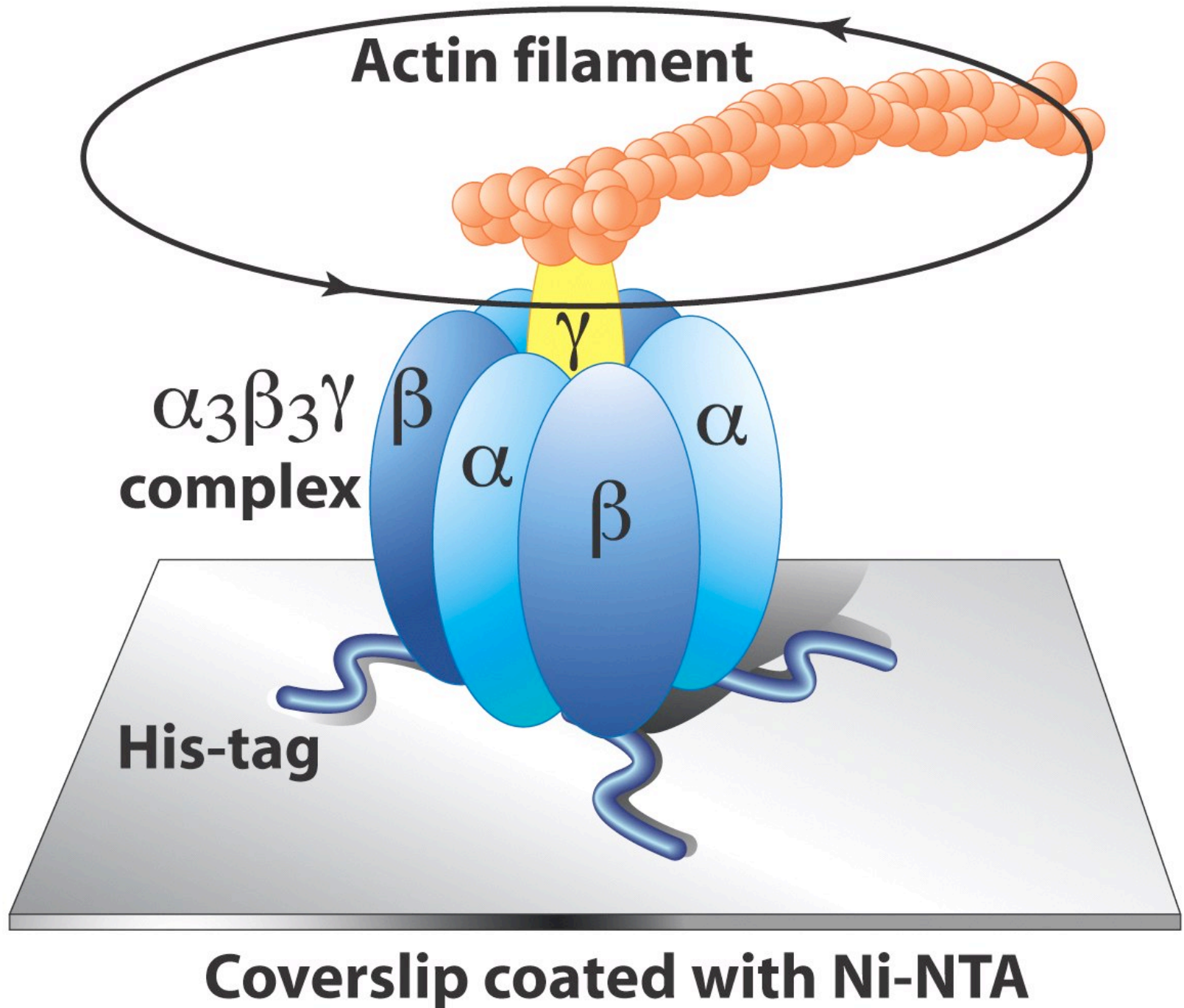


Figure 5-28 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)









**NOBEL PRIZES**

ALFRED NOBEL

PRIZE AWARDEES

NOMINATION

PRIZE ANNOUNCEMENTS

AWARD CEREMONIES

By Year

Nobel Prize in Physics

**Nobel Prize in Chemistry**

Nobel Prize in Medicine

Nobel Prize in Literature

Nobel Peace

Chemistry



## The Nobel Prize in Chemistry 1978

"for his contribution to the understanding of biological energy transfer through the formulation of the chemiosmotic theory"

**Peter D. Mitchell**

United Kingdom

Glynn Research  
Laboratories  
Bodmin, United Kingdomb. 1920  
d. 1992

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Comments &amp; Questions



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**The 1978 Prize in:**

Chemistry

⏮ Prev. year

Next year ⏭

**The Nobel Prize in  
Chemistry 1978**

Press Release

Presentation Speech

**Peter Mitchell**

Biography

Nobel Lecture

Banquet Speech

By Year

Nobel Prize in Physics

**Nobel Prize in Chemistry**

Nobel Prize in Medicine

Nobel Prize in Literature

Nobel Peace Prize

Prize in Economics



## The Nobel Prize in Chemistry 1997

"for their elucidation of the enzymatic mechanism underlying the synthesis of adenosine triphosphate (ATP)"



**Paul D. Boyer**

🕒 1/4 of the prize

USA

University of California  
Los Angeles, CA, USA

b. 1918



**John E. Walker**

🕒 1/4 of the prize

United Kingdom

MRC Laboratory of  
Molecular Biology  
Cambridge, United  
Kingdom

b. 1941

"for the first discovery of an ion-transporting enzyme, Na<sup>+</sup>, K<sup>+</sup> - ATPase"



**Jens C. Skou**

🕒 1/2 of the prize

Denmark

Aarhus University  
Aarhus, Denmark

b. 1918



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Comments & Questions



Tell a Friend

### The 1997 Prize in:

Chemistry

⏮ Prev. year

Next year ⏭

### The Nobel Prize in Chemistry 1997

Press Release

Presentation Speech

Illustrated Presentation

#### Paul D. Boyer

Autobiography

Nobel Lecture

Interview

Nobel Diploma

Photo Gallery

Other Resources

#### John E. Walker

Autobiography

Nobel Lecture

Nobel Diploma

Photo Gallery

Other Resources

#### Jens C. Skou

Autobiography

Nobel Lecture

Nobel Diploma

Photo Gallery

Banquet Speech

Other Resources



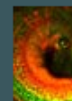
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year's Nobel  
Laureates



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News from  
Nobelprize.org



Oldest,  
youngest,  
most awarded



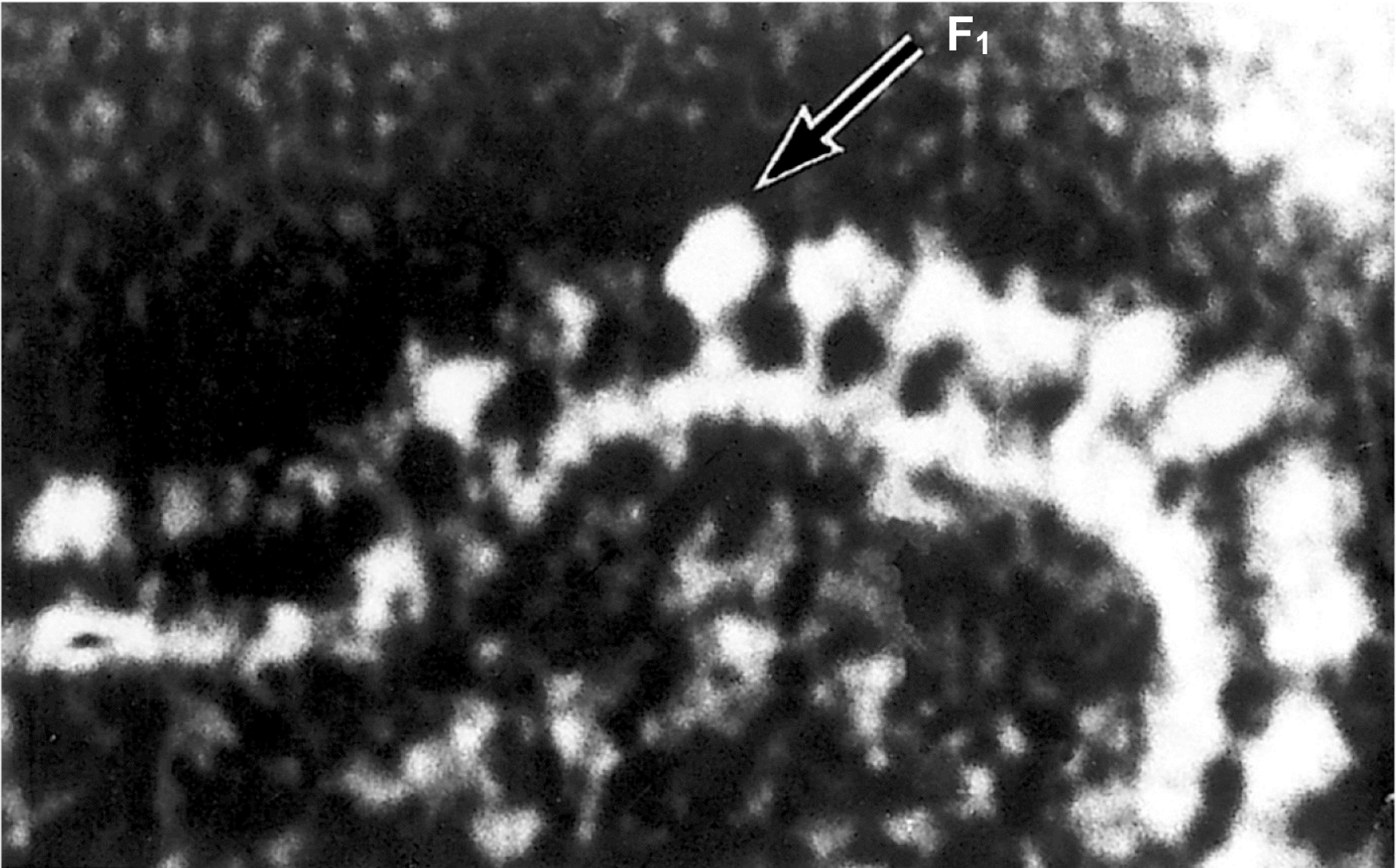
How do we  
hear? Play and  
learn.



Memories of  
that call



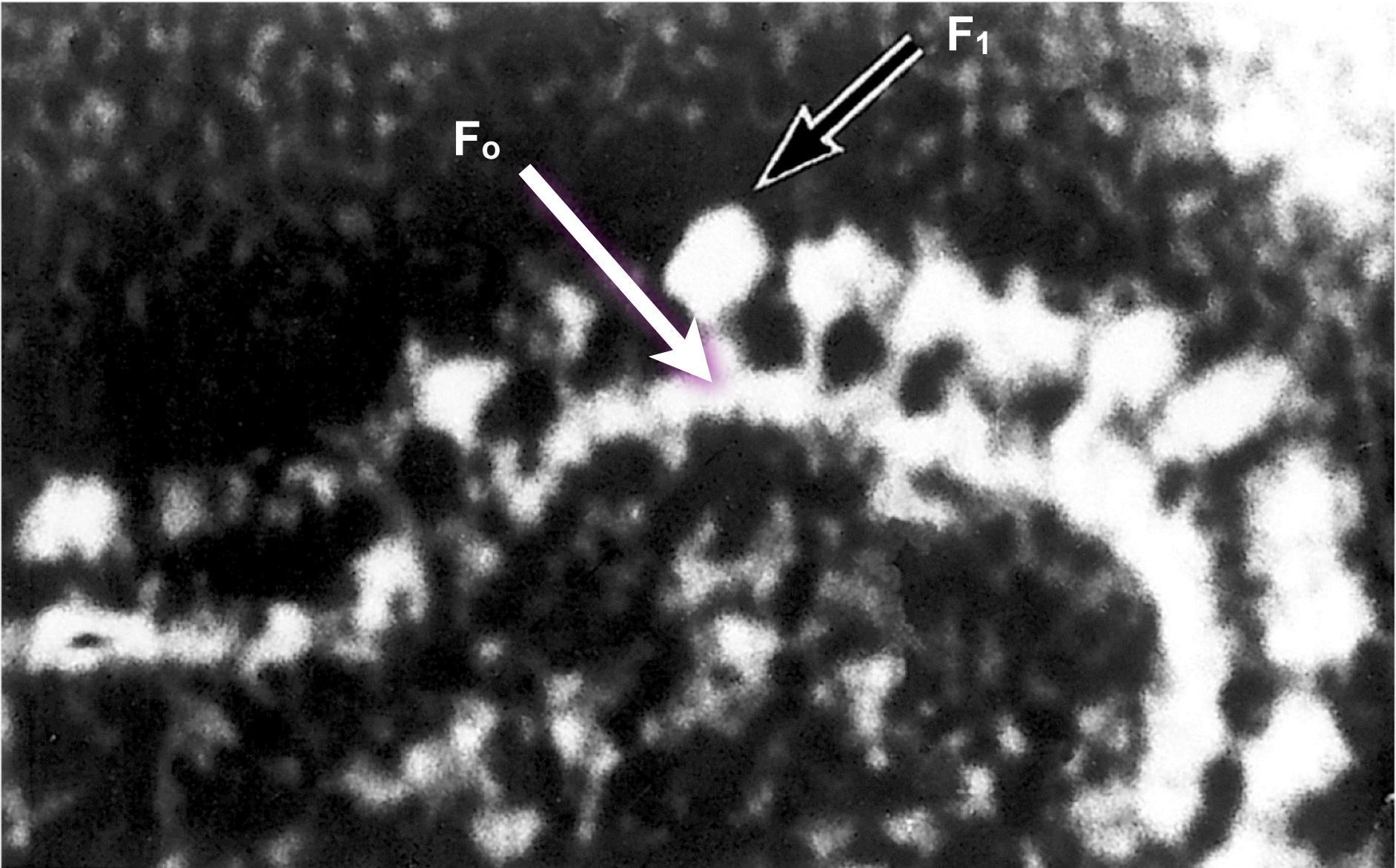
$F_o$



10 nm

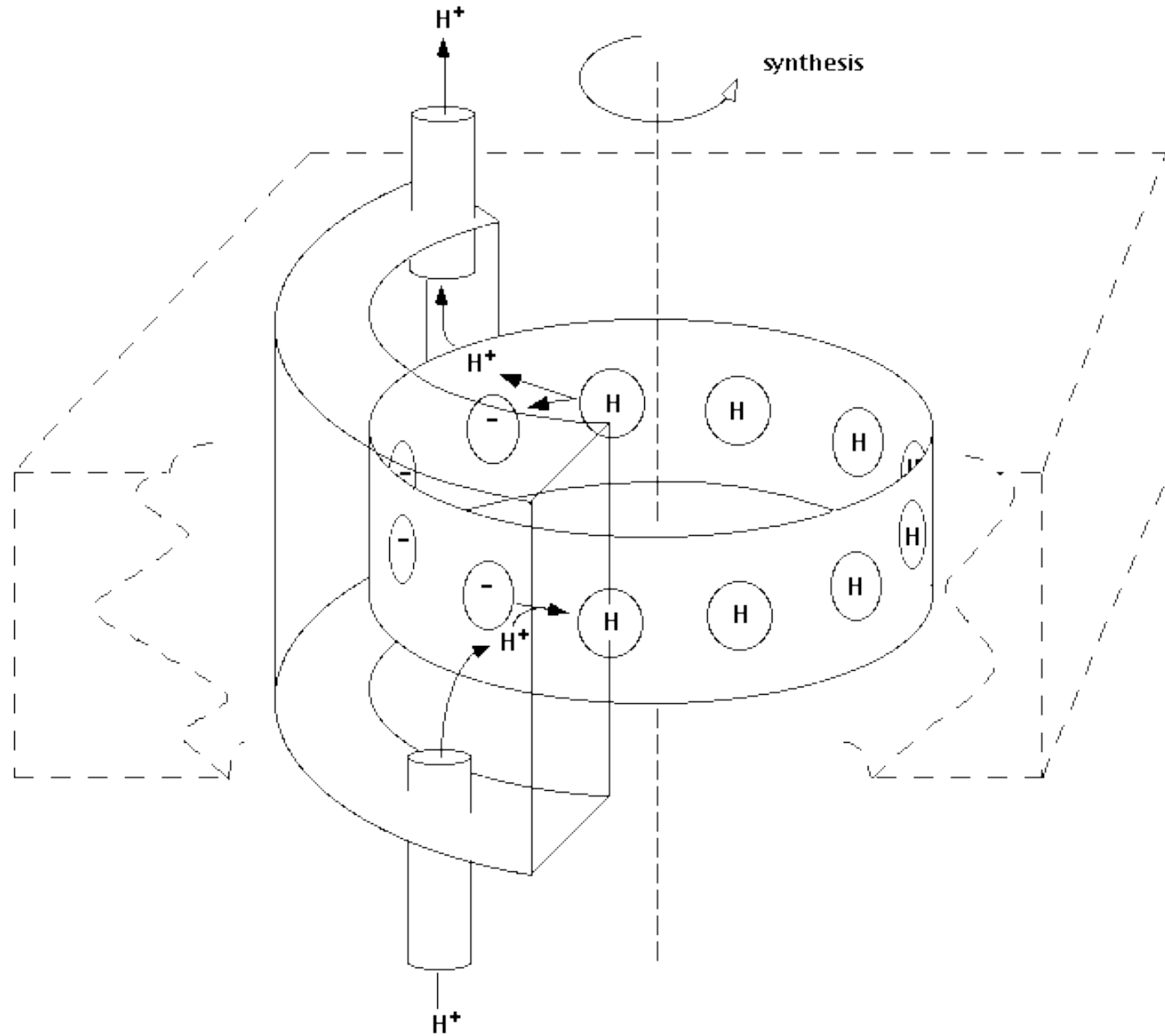
Figure 5-21 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)



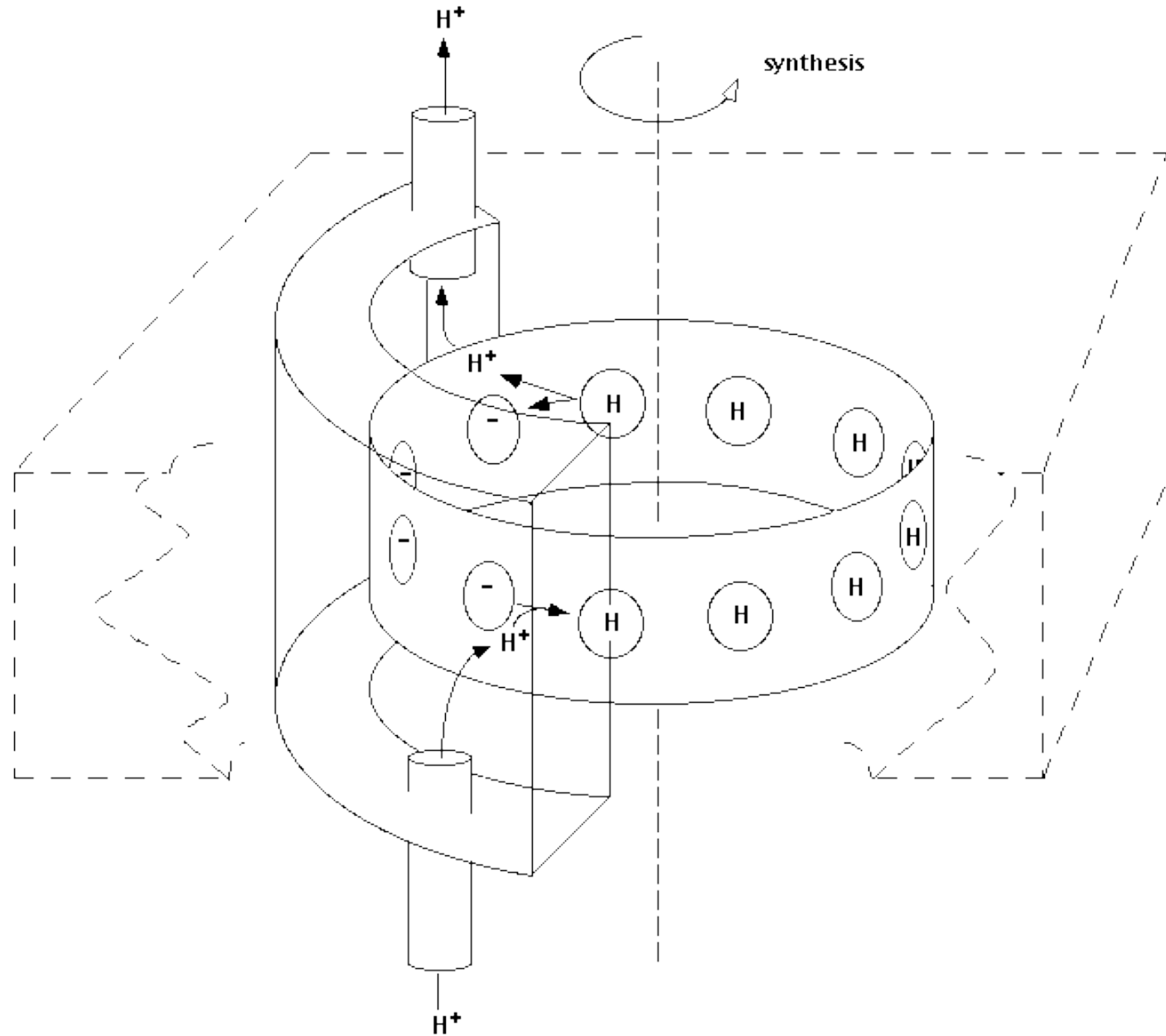


10 nm

Figure 5-21 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)



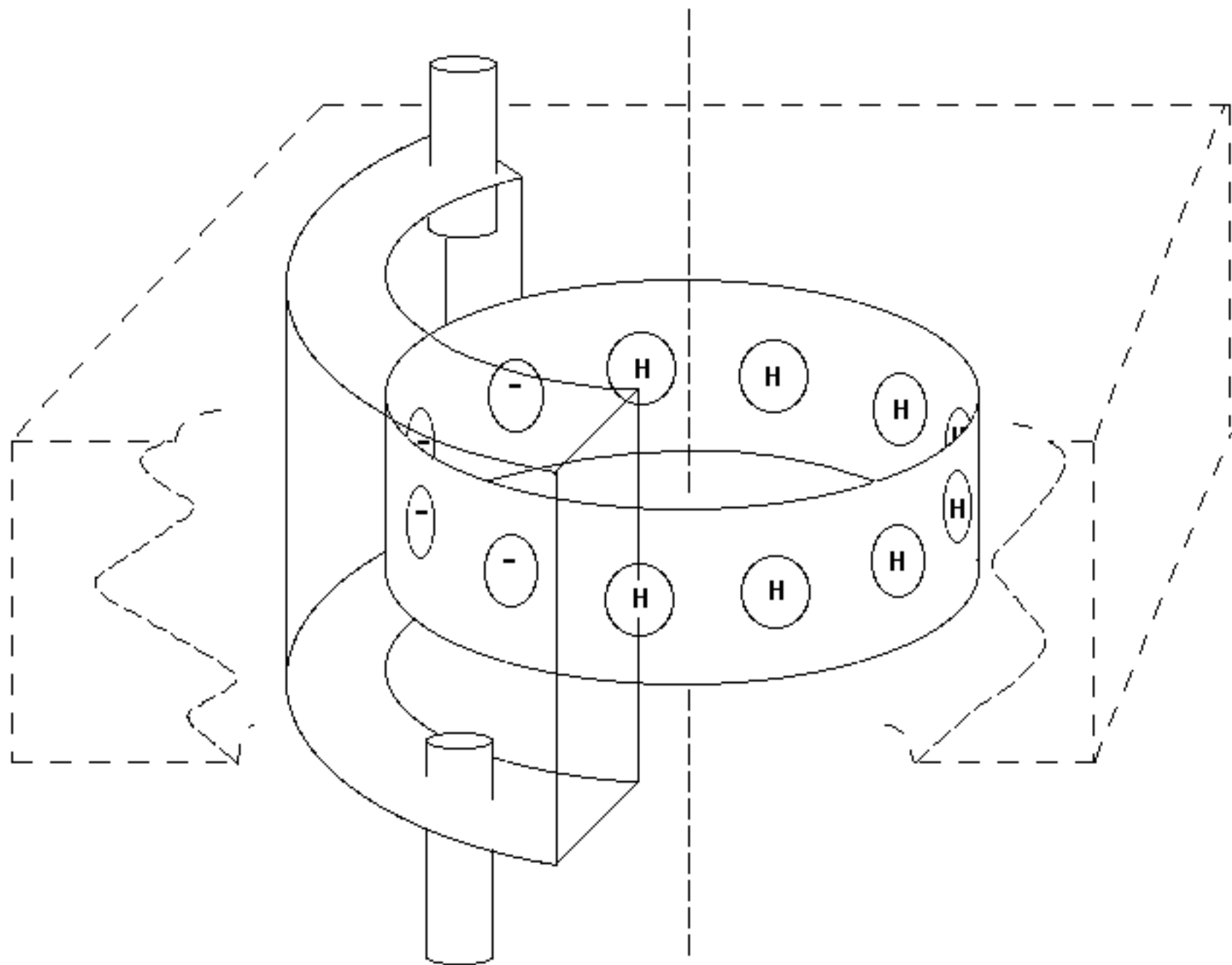
Rotation of the Fo-ATPase. Fo-ATPase as a proton-driven, rotary stepping motor, as proposed by Junge (1997).

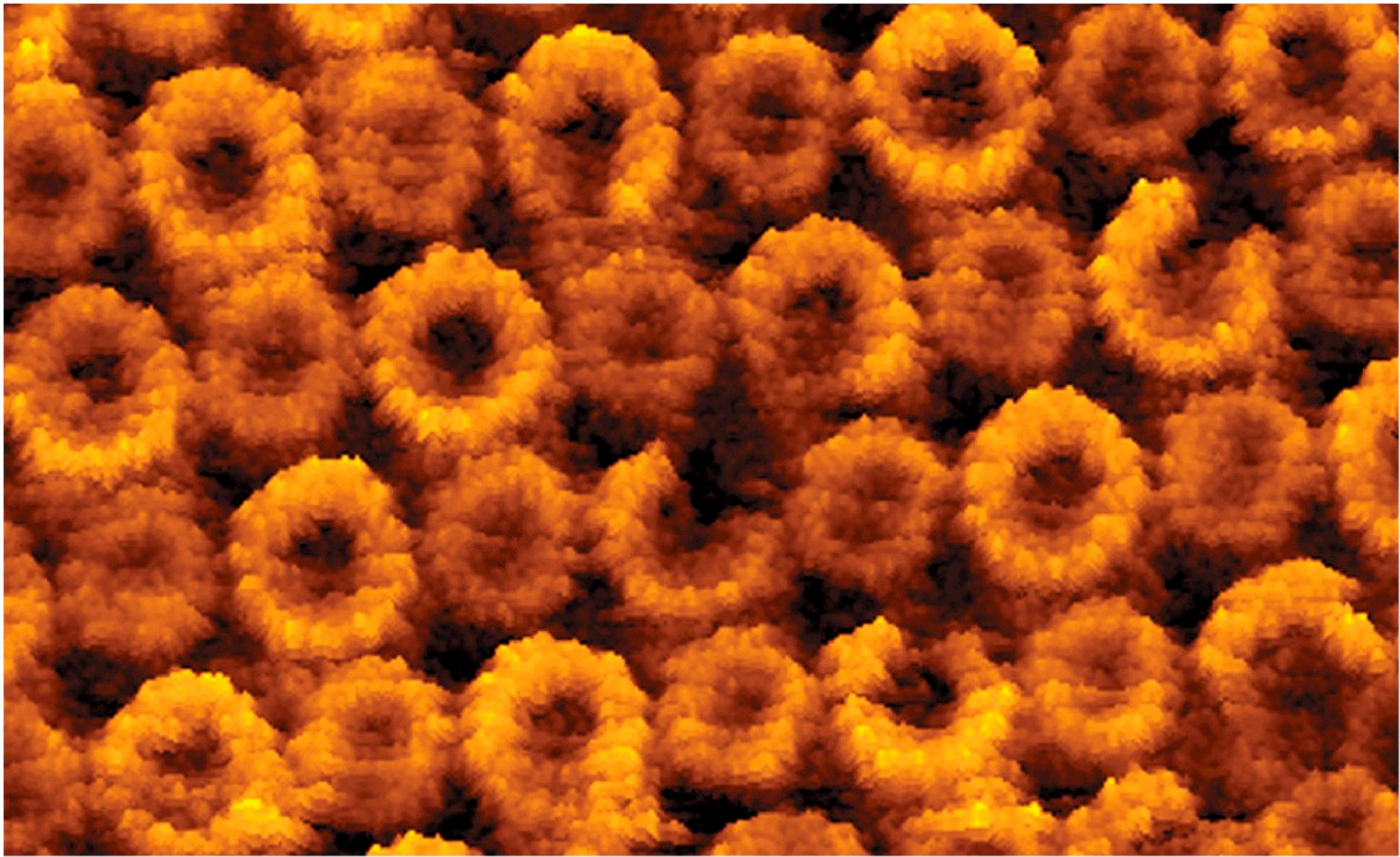


Rotation of the Fo-ATPase. Fo-ATPase as a proton-driven, rotary stepping motor, as proposed by Junge (1997).

<http://jfa.sbcs.qmul.ac.uk/~john/webstar/ltn/06/3ATP.html>







5 nm

Figure 5-24a Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)



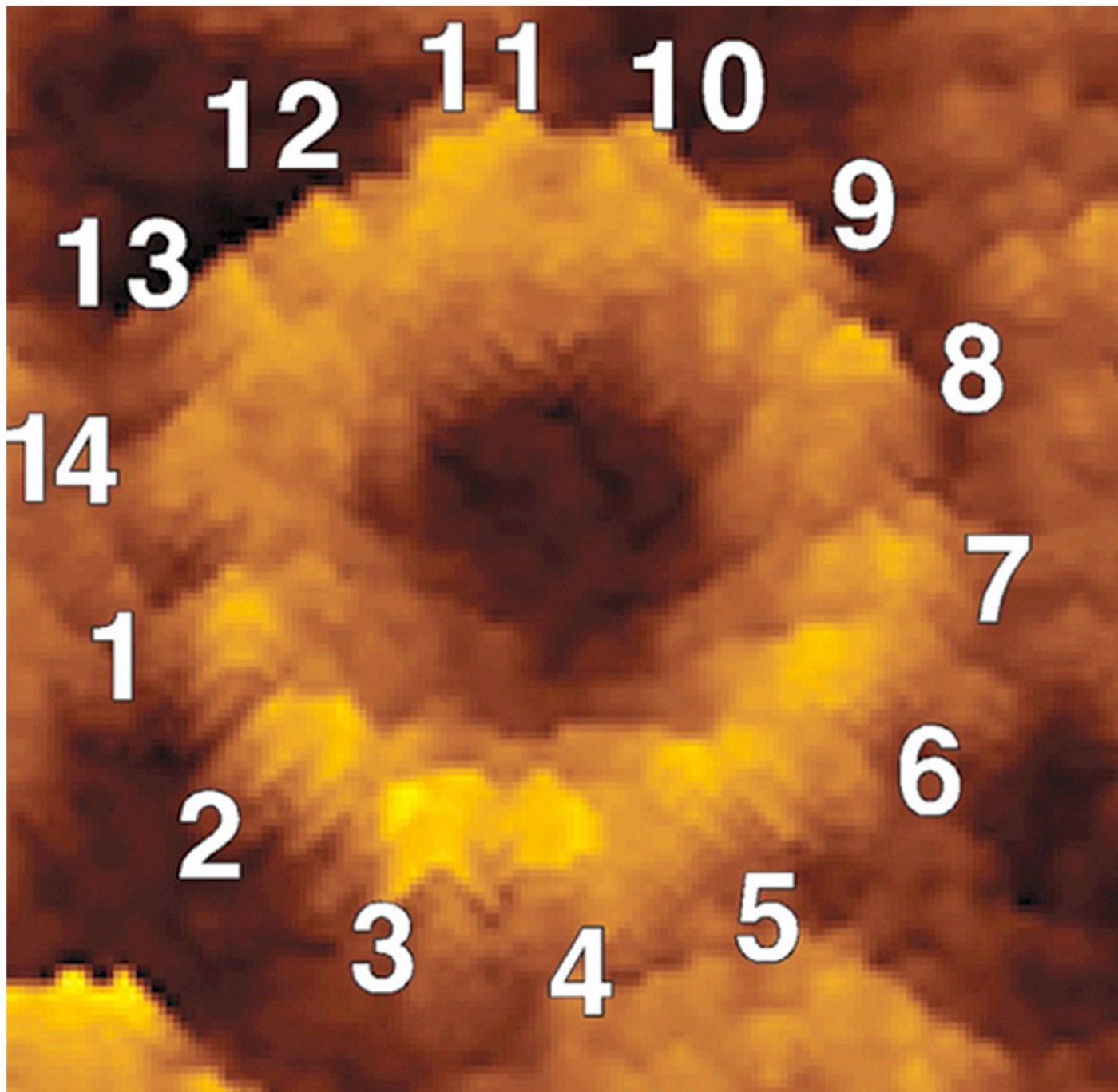


Figure 5-24b Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)

# The $F_1$ - $F_0$ ATPase

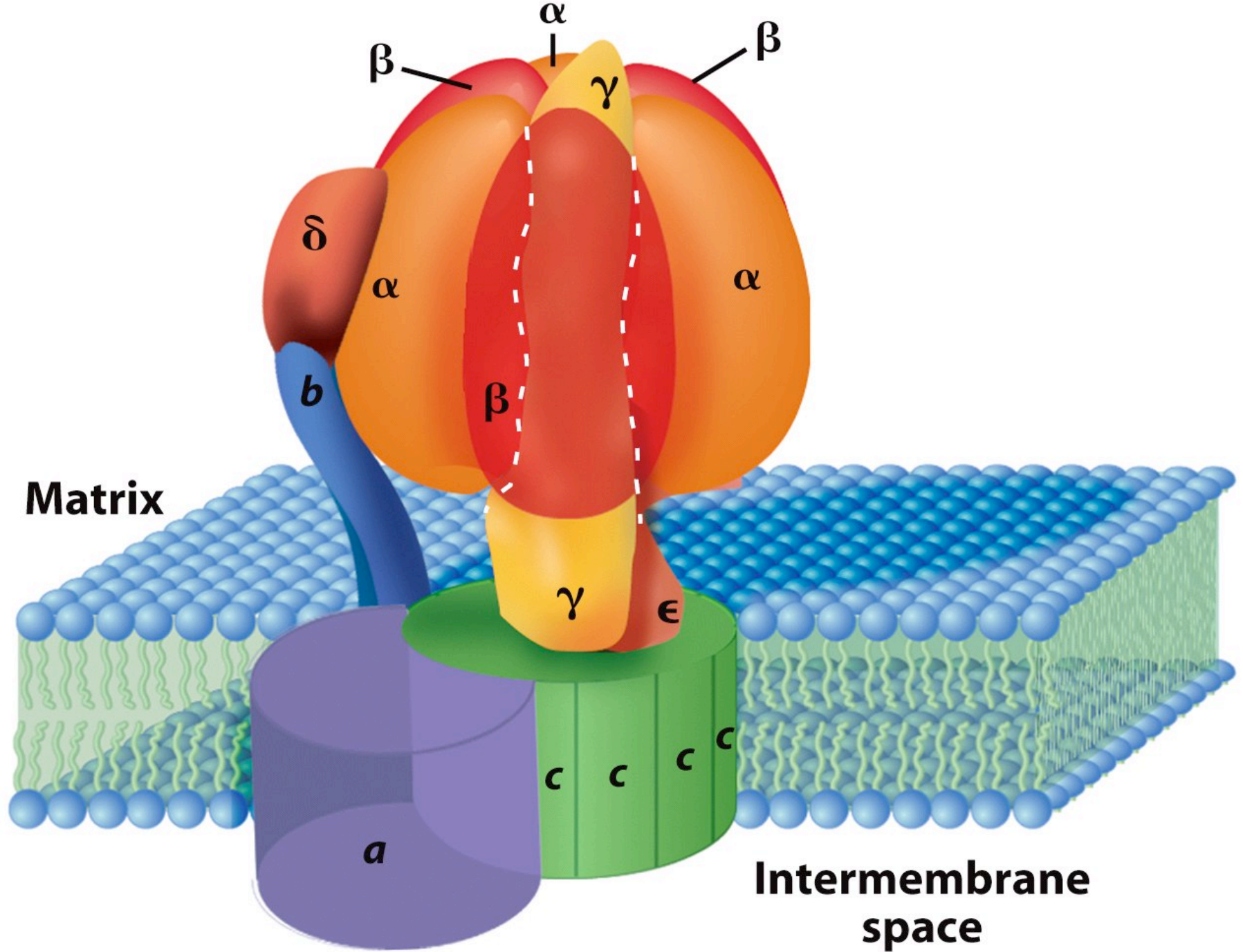


Figure 5-23b Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)



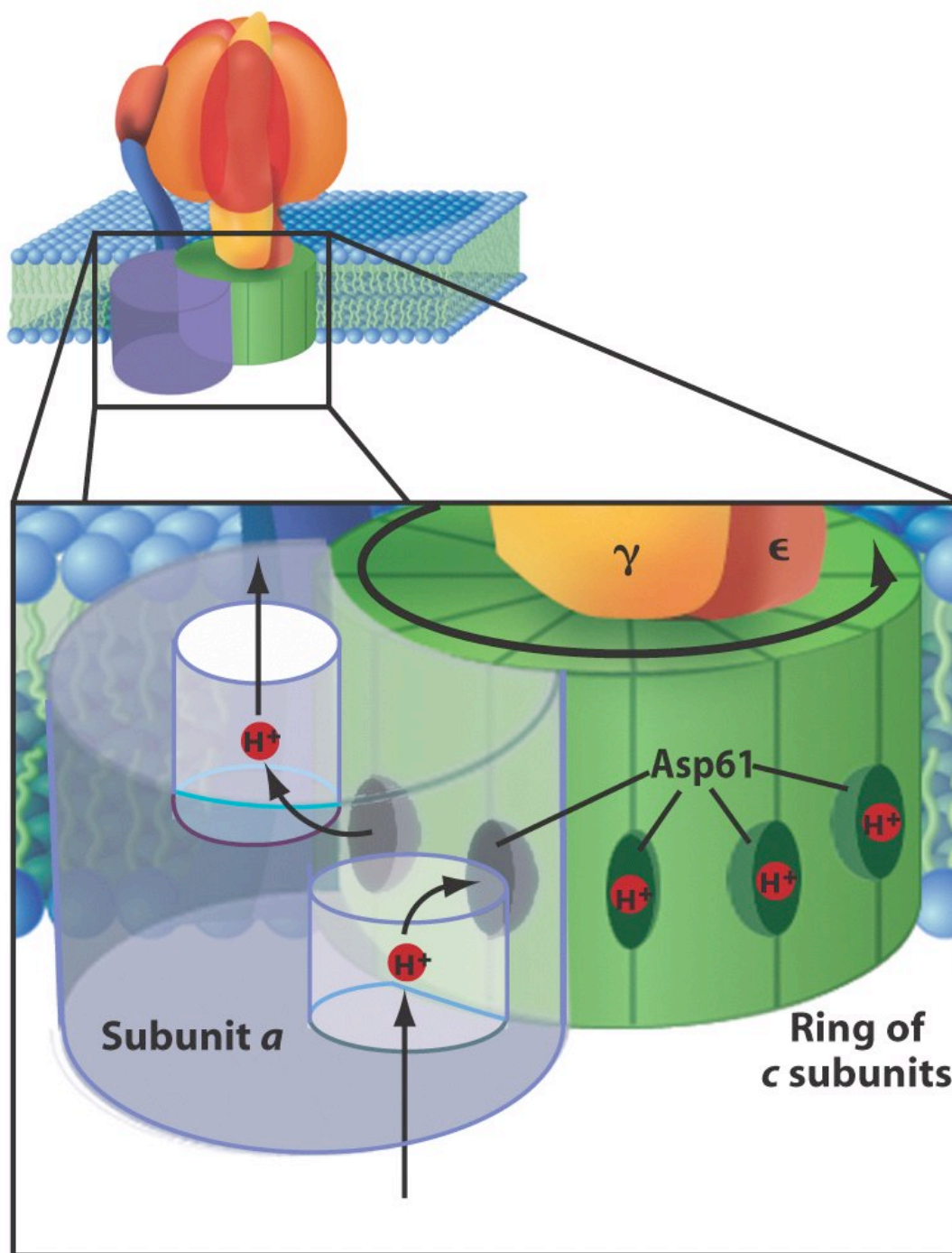


Figure 5-29 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)

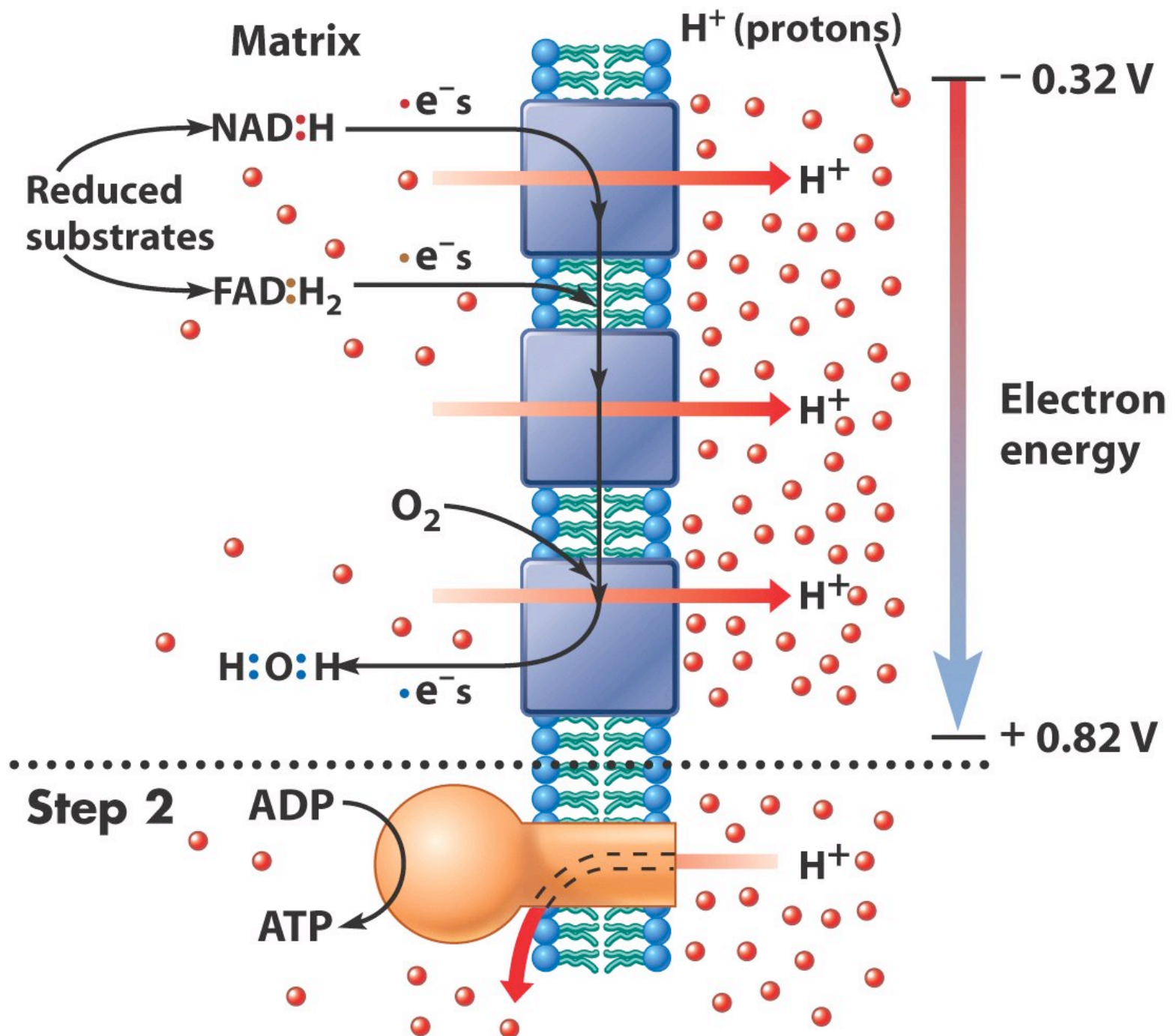


Figure 5-10 Cell and Molecular Biology, 4/e (© 2005 John Wiley & Sons)





# Movies

These movies were created by Said Sannuga in collaboration with John Walker and Andrew Leslie.

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- View from above and then below the  $F_1$  domain along the rotating  $\gamma$ -subunit

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- How the rotating  $\gamma$ -subunit imposes the conformational states on a  $\beta$  subunit required for substrate binding, ATP formation and ATP release

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- Three conformations of a catalytic  $\beta$ -subunit produced by  $120^\circ$  rotations of the central  $\gamma$ -subunit

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- Changes in the positions of sidechains in the catalytic site of  $F_1$ -ATPase bringing about binding and subsequent hydrolysis of ATP

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- View from above and then below the  $F_1$  domain along the rotating  $\gamma$ -subunit
- How the rotating  $\gamma$ -subunit imposes the conformational states on a  $\beta$  subunit required for substrate binding, ATP formation and ATP release
- Three conformations of a catalytic  $\beta$ -subunit produced by  $120^\circ$  rotations of the central  $\gamma$ -subunit
- Changes in the positions of sidechains in the catalytic site of  $F_1$ -ATPase bringing about binding and subsequent hydrolysis of ATP
- The rotary mechanism of mitochondrial ATP synthase

View from above and then below the  $F_1$  domain  
along the rotating  $\gamma$ -subunit

© Medical Research Council



How the rotating  $\gamma$ -subunit imposes conformational states on a  $\beta$ -subunit required for substrate binding, ATP formation and ATP release.

© Medical Research Council

MRC

Mitochondrial  
Biology Unit



# Three conformations of a catalytic $\beta$ -subunit produced by 120° rotations of the central $\gamma$ -subunit

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MRC

Mitochondrial  
Biology Unit

Changes in the positions of side-chains in the catalytic site of  $F_1$ -ATPase bringing about binding & subsequent hydrolysis of ATP.

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MRC

Mitochondrial  
Biology Unit

# The rotary catalytic mechanism of mitochondrial ATP synthase.

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Mitochondrial  
Biology Unit







***Thank you for listening***

# Membrane Biochemistry

Lectures by

**John F. Allen**

School of Biological and Chemical Sciences, Queen Mary, University of London

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